THESIS

THE INFLUENCE OF HERBICIDES

ON

MICROBIAL ACTIVITY IN THE SOIL

Submitted by Donald B. Bakes

In partial fulfillment of the requirements

for the Degree of Master of Science

Colorado

Agricultural and Mechanical College

Fort Collins, Colorado

July, 1956

COLORADO A. & M. COLLEGE
FORT COLLINS COLORADO

COLORADO AGRICULTURAL AND MECHANICAL COLLEGE

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378.788
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WE HEREBY RECOMMEND THAT THE THESIS PREPARED UNDER OUR
SUPERVISION BY DONALD B. BAKES
ENTITLED THE INFLUENCE OF HERBICIDES ON MICROBIAL
ACTIVITY IN THE SOIL
BE ACCEPTED AS FULFILLING THIS PART OF THE REQUIREMENTS FOR THE
DEGREE OF MASTER OF SCIENCE.
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ACKNOWLEDGMENT

The author wishes to express his sincere gratitude to all who gave their time and effort in assisting in this investigation.

The author is especially indebted to Professor Robert Kunkel under whom this study was initiated and for his valuable assistance on the details of this work. To Professor Donal D. Johnson in guiding this study to completion and the preparation of the manuscript I express my sincere appreciation. Also to Professors S. M. Morrison and E. B. Crone for their suggestions in the preparation of the manuscript.

The author wishes to thank Professor E. E. Remmenga for his suggestions on the statistical techniques, and to Doris Anderson who helped with the statistical analysis.

In addition the author wishes to express his appreciation to the American Cyanamid, Balcom and Geigy Chemical companies for supplying the chemicals used in this study.

A special acknowledgment is extended to the American Cyanamid Company for the research fellowship grant which made this study possible, and to the Colorado A & M Research Foundation for the special grant which made it possible to bring this investigation to completion.

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INTRODUCTION

Present day trends in agriculture indicate that farmers must out production costs if they are to realize a profit. For example, to produce a crop of field onions in Colorado four to five hand weedings are required. may cost as much as one hundred dollars per acre. use of herbicides is one effective way of cutting production costs. Much empirical data are available on the effects of herbicide applications on weed control. type of data answers many of the questions raised by growers such as type of herbicide to use, rates and dates of applications and whether it will increase or decrease yields. Previous investigations in this department (22) have answered some of these questions; however other fundamental questions remain unanswered. Two examples follow: does the application of herbicides to the soil alter the exchange complex by the replacement of an ion adsorbed on the soil colloid with an herbicide ion? possible effects do herbicides have on soil microorganisms?

Fundamental information is necessary if we are to eliminate some of the hazards which accompany the use of

agricultural chemicals. The scope of this study will be limited to that portion of the problem which involves the production of nitrate nitrogen by microorganisms in the soil as it is influenced by temperature, soil type, kind and rate of herbicide application as a function of time.

The nature of the biological medium into which herbicides are placed needs to be clearly understood before intelligent cultural practices can be utilized efficiently. The soil is a dynamic heterogeneous medium consisting of very many different kinds and numbers of organisms, the numbers and kinds depending upon the soil environment and man's treatment. A system of checks and balances imposed by physical, chemical, and biological variations within the soil serves to decide the kind and number of the active population at a given time. Thus the application of an herbicide to the soil would change the environment and alter the biological composition of the soil. This poses the question: is the magnitude of the change brought about by herbicides enough to alter the nitrate nitrogen produced by microorganisms? One important group of microorganisms is the chemosynthetic autotrophs which change the ammonia nitrogen produced by the heterotrophs from the soil organic matter, into nitrate nitrogen. This group is sensitive to change in the soil environment and hence the nitrate produced by them

may be used as an indication of the effect of a given herbicide on one segment of the soil microbial population. Since all higher plants, including onions, are dependent on the production of nitrate nitrogen, it would be desirable to know whether or not the herbicides may in any way effect this fundamental process.

If the herbicides used in weed control on onions affect the nitrate nitrogen producing bacteria, then the value of the herbicide must be correlated with the changes in amounts of nitrate nitrogen. For this reason the investigation was undertaken.

Problem

Do herbicides influence microbial activity in the soil?

<u>Problem analysis.</u>—In order to answer the main problem it was found necessary to obtain answers to the following four questions:

- 1. What is the effect of herbicides on microbial activity when applied to the soil?
- 2. What is the effect of rate of application of herbicides on microbial activity as measured by production of nitrate nitrogen?
- 3. What is the duration of the effect of herbicides as measured by microbial activity?

4. What is the duration of the effect of herbicides on microbial activity as measured by nitrate nitrogen?

Delimitations. --This investigation has been limited to (1) three soil types: Gilcrest sand, Rocky Ford loam and Las Animas clay; (2) three herbicides: Calcium cyanamide (CaCN2), o-isopropyl n-(3 chlorophenyl) carbamate (chloro IPC) and 2 chloro 4,6-bis (diethylamino)-s triazine (444E); (3) three soil temperatures: 13°, 25°, and 35° Centigrade; (4) four dates of sampling: 5, 10, 20, and 30 days after the start of incubation; (5) microbial activity to be measured by changes in nitrate nitrogen; (6) four herbicide rates.

REVIEW OF LITERATURE

Herbicide residues in soils have received much less consideration than the effects of their applications to plants. This placement of emphasis can be attributed to the readily observable effect of the herbicide on plant growth, whereas the effects of herbicide residues in soils may not be readily apparent. Studies of herbicides in soils have dealt primarily with the persistence of the herbicide as it may effect subsequent crops and with the effect of herbicides on microorganisms.

Effects of 2,4-dichlorophenoxyacetic acid on soil microorganisms have been investigated. Koike and Gainey (12) state 2,4-D is non-toxic to most microorganisms at concentrations used in the field for weed control. Fults and Payne (10) and other investigators (5,11,17,34, 25) report inhibitory effects on organisms by 2,4-D at concentrations used for weed control.

Aldrich (1) states that all microorganisms are not affected to the same degree by a particular herbicide, just as all higher plants do not react the same. He also states that the selective action of 2,4-D on microorganism

population may change with repeated herbicide application. The rate of 2,4-D decomposition in soil indicates slight decomposition immediately after treatment, followed by rapid decomposition. Presumably a lag period is necessary for build-up of organisms capable of breaking down 2,4-D, indicating that the number of such organisms is limited at a given time. Audus (3) and Newman (17) working with 2,4-D and 2-methyl-4-chlorophenoxyacetic acid (MCP) report that pretreatment with these compounds results in more rapid decomposition of subsequent applications of the compounds. This suggests an accumulation of organisms capable of using such compounds as a source of carbon. However, Audus (2) was not able to culture the organism responsible for 2.4-D decomposition in the absence of 2,4-Therefore, he concludes that 2,4-D is the only source of carbon utilizable by these organisms.

Norman and Newman (21) state that the build-up of organisms capable of decomposing a particular herbicide may reduce the effectiveness of subsequent soil applications of the herbicide.

Aldrich (1) states that herbicides may be removed from the soil by one or more of the following ways:

(1) leaching, (2) fixation by the soil colloids, (3) decomposition, and (4) volatilization. Conditions that affect these means of removal determine the persistence of an herbicide in the soil.

It appears that decomposition of herbicides in the soil is accomplished through the action of soil organisms. A number of investigations (2,3,8,9,13,14, 15,18,19,20,23,28,30) have shown that soil moisture, soil temperature, organic matter content, and other factors most conducive to microbial activity are also most conducive to herbicide decomposition. ZoBell (31) states that aliphatic hydrocarbons are exidized more readily than aromatic or naphthenic compounds. Within certain limits, long-chain hydrocarbons are attacked more readily than similar compounds of small molecular weight. The addition of aliphatic side chains increases the susceptibility of cyclic compounds to microbial attack.

The reports of decomposition of herbicides by microorganisms indicate that small differences in chemical structure have a pronounced effect on decomposition by soil organisms. Isopropyl N-(3-chlorophenyl)-carbamate is readily attacked by microorganisms in soil (19,23,27), while 3-p-chlorophenyl-l,l-dimethylurea is extremely resistant to attack.

Audus (2) found that the organisms capable of breaking down 2,4-D were able to breakdown 2 methyl-4-chlorophenoxyacetic acid at a lower rate. The organism involved in 2,4-D decomposition did not affect decomposition of 2,4,5-T (2,4,5-trichlorophenoxyacetic acid).

Early workers believed that microorganisms were responsible for the change from cyanamide to ammonia in the soil. Crowther and Richardson (7) state that although some bacteria and fungi are able to attack cyanamide, many inorganic catalysts for the cyanamide to urea reaction were discovered by Ulpiani, Kappen, and others. Cowie (6) showed that in sterile soils urea was rapidly formed and accumulated; in partially sterilized soils the urea disappeared again after a few days and ammonia and nitrates accumulated.

Temme (29) found that the speed with which the cyanamide in the soil is being converted appears to be connected with the humidity of the soil and the temperature whereby the conversion takes place. He concluded that the conversion of calcium cyanamide to urea is accomplished by physico-chemical means, while the conversion of urea to nitrate nitrogen is accomplished biologically.

Mukerji (16) measuring the microbiological activity of the soil treated with calcium cyanamide found no relationship between bacterial numbers as determined by plate counts and the changes in the amounts of simple nitrogen compounds. He states that since the plating method is highly selective, the rate of carbon dioxide production was used as a general integration of microbiological activity. The results of Mukerji's investigation show

that in two different experiments calcium cyanamide when compared with ammonium sulfate increased the carbon dioxide evolution 18 per cent.

METHODS AND MATERIALS

One hundred pound samples of the three soil types were collected in the field and brought to Fort Collins. These samples were air-dried then sieved, the soil passing through a 2 mm. sieve was reserved for the study.

A tension equal to 70 cm. of water (near ideal moisture content) for the Gilcrest sand from Weld county, was 1.55 ml. per 25 g. of soil; for the Rocky Ford loam from Otero county, 6.55 ml. per 25 g. of soil; and for the Las Animas clay from Otero county, 7.14 ml. per 25 g. of soil. Other properties of these soils are given in Table 1.

Table 1 .-- COOPERATIVE SOILS LABORATORY ANALYSIS REPORT.

Soil class	Depth	pH 1:5	Per cent soluble salts	Per cent 0,E.	Per cent lime	1b./A. P205	1b./A. K20
Sand	0-6"	6.8	0.07	1.8	0.0	380	39 7
Loam	0-6"	7.9	0.13	1.4	7.7	47	370
Clay	0-6"	8.0	0.16	2.4	13.1	96	689

Twenty-five gram samples of the air-dry soil were weighed and placed into 25 ml. erlenmeyer flasks. The herbicides used in this experiment were added to these flasks at rates equivalent to rates used in field applications as indicated by Table 2.

Table 2. -- HERBICIDE RATES AS EXPRESSED IN POUNDS, KILO-GRAMS, AND MICROGRAMS.

Herbicide	Pounds per acre	Kilo g. per 25 g. soil	Micrograms per 25 g.soil
CIPC	3	1.36	3 7.5
CIPO	3 / 3	2.72	75.0
CIPC	6	2.72	75.0
444E	10	4.54	97.5
444E	15	6.80	146.2
444E	20	9.07	195,0
Cacna	100	45.36	1250.0
CaON2	200	90,72	2500.0
CaCN	400	181.44	5000.0

The three herbicides used in this experiment were calcium cyanamide (CaCN₂), Chloro-IPC or CIPC (o-isopropyl n- (3 chlorophenyl) carbamate, and 444E (4,6-bis (diethylamino) -s triazine. The structural formula for each herbicide is presented.

$$Ca = N - C \equiv N$$

Calcium cyanamide Chloro-IPC CH_3
 CH_3

The materials soluble in water (CIPC) and 444E were dissolved in water in such an amount that the water required to bring the soil to a water content equal to that at 70 centimeters of water tension would contain the proper amount of herbicide. The solutions were applied by means of a burette. The insoluble calcium cyanamide was mixed with a diluent of quartz sand to pass a 200 mesh sieve. The herbicide-diluent mixture equivalent to recommended rates for weed control in onions was weighed and mixed with each soil sample. Water in proper amount was then added to these samples.

Three incubation temperatures were used: 13°C., 25°C., and 35°C. These temperatures were maintained in an insulated chamber by means of thermostatically controlled infrared heat lamps. A high relative humidity was maintained in the chambers by a pan of water which was just slightly smaller than the dimensions of the bottom of the box. Temperature and humidity were recorded on a weekly hydro-thermograph. All erlenmeyer flasks were stoppered with one-holed rubber stoppers to minimize changes in moisture content.

One hundred and forty-four erlenmeyer flasks containing 25 grams of soil plus the herbicide treatment were placed in three different temperature chambers. One-fourth of all the samples from each chamber were removed at the end of 5, 10, 20, and 30 days. Upon removal from the chambers, all samples were immediately oven-dried and stored prior to analysis.

Explanation of chemical analysis

- 1. Nitrate nitrogen was removed from each 25 g. soil sample by extraction with an 80 ml. solution of silver sulfate and copper sulfate (2 g. AgSO₄ 30 g. CuSO₄ per 12 liters distilled water).
 - 2. Shake 10 minutes.
- 3. Add about 0.6 g. precipitating mixture (10 parts MgCO3 and 4 parts Ca(OH)2).

- 4. Shake 10 minutes.
- 5. Filter (Whatman 42) and catch extract in 50 ml. erlenmeyer flasks.
 - 6. Pipet aliquot of 2 ml. into 100 beaker.
 - 7. Dry aliquot in oven (80 to 85°C).
- 8. Add 2 ml, phenoldisulfonic acid reagent (dissolve 25 g, pure white phenol in 150 ml, concentrated H₂SO₄, add 75 ml, fuming H₂SO₄, 13 to 15 per cent SO₃. Heat at 100°C, for two hours.
- 9. Allow to stand for 10 minutes, making sure that all residue in the beaker has been contacted.
 - 10. Add 16 ml. of distilled water.
 - 11. Add 20 ml. of 1:1 NH40H.
- 12. Cool and read per cent transmission using a Cenco Photelometer and a filter of 410 to 440 mu.

 The transmission data was converted to micrograms of nitrate nitrogen from a calibration curve, and the data is recorded in appendices I and II.

STATISTICAL ANALYSIS OF DATA

Biological data is highly variable and well designed experiments always provide a means of assessing the magnitude of this variation. Conventionally, replicates of each treatment accomplish this objective with two replicates being the minimum and a much larger number being more desirable. The major limitation of this technique is that large numbers of samples must be processed. In the present experiment 432 individual analyses were made and the experiment had but one replication. The necessary precision was achieved by using a statistical technique for analyzing the data. This technique is described in the succeeding paragraph.

evaluate an experiment of this type without the use of replicates is based on the fact that when an analysis of variance is interpreted it is impossible, to a large degree, to interpret third and higher order interactions. These higher order interactions are lumped with the error term and as a result increase the overall and individual item precision accordingly. In the experiment described five main effects were considered; temperature, soil type,

rate of herbicide, kind of herbicide, and sampling dates:

10 first order interactions, 10 second order interactions;

third and higher order interactions were lumped with the

error term.

The statistical technique by which the data were handled is given in detail in appendices I and II. The data were analyzed in two different manners with identical results. A 33 x 24 factorial design was used in the experiment and the production of nitrate nitrogen as influenced by time, temperature, soil type, rate of herbicide, kind of herbicide and their interactions were measured.

In the succeeding discussion it is to be understood that whenever the term nitrate nitrogen is used it will refer to the average amount of nitrate nitrogen produced.

The analysis of the data shows that a significant increase in nitrate was obtained as a result of the action of all of the major variables under consideration. The effect of temperature on nitrate production was linear; the higher the temperature, the higher the nitrate production (see Table 4). Since the temperatures used were well within the range normally found in these soils, the data confirm predictions which could have been made from theory.

The effect of soil type was very significant. The data show that the soils studied nitrify at different and characteristic rates. The major factor involved in the differences between the soils appears to be the amount of clay which the soils contain. The Gilcrest sand produced less nitrate than the Rocky Ford loam which in turn produced less than the Las Animas clay. Microbial populations also increase with the clay content of the soil and hence one explanation of this effect of soil type may be that a higher number of nitrifying bacteria is associated with the increase in clay content (see Tables 6 and 7).

Nitrate nitrogen increased in all of the soils as a function of time. This is as would be expected and is a direct result of the build up and adaptation of the microbial population.

The main effect of herbicides indicates that there is no statistical difference in the amount of nitrate nitrogen produced by soils treated with either Chloro-IPC or 444E. Soils treated with these materials however, produced significantly less nitrate nitrogen than those to which calcium cyanamide had been added. There was no difference in the amount of nitrate nitrogen produced by soils treated with Chloro-IPC or 444E and the nontreated soil.

when the effect of herbicide rates were compared, it was noted that there was no statistical difference between the untreated control, and the first rates. There was no statistical difference between the second and third rates, however, the amount of nitrate nitrogen produced by both the second and third rates was significantly higher than in the nontreated soil.

An examination of the first and second order interactions showed that with the exception of herbicide by herbicide rate, and herbicide by sampling date, none of the interactions showed changes in effect. That is, the soils studied did not act differently at different temperatures, times or with different herbicides or herbicide rates. Some of the more important interactions are shown in Tables 4, 5, 6, 7, 8 and Figs. 1 and 2.

Table 3 shows the relationship between temperature and herbicide. The amount of nitrate nitrogen produced in soils treated with 444E and Chloro-IPC increased with temperature. The difference between the two herbicides was slight but significant at 25° and 35°C.

The relation between rate of herbicide application and temperature is shown in Table 4. This data shows that there was no significant increase in nitrate nitrogen production at any temperature when the untreated soil and the first rate are compared. The data in Table 4 also

indicates that there was a significant increase in nitrate nitrogen production over the check at all temperatures and at all other herbicide rates.

The relationship between kind of herbicide and soil type as these variables affect nitrate accumulation is shown in Table 5. This data indicates that the herbicides did not act differently in the soils studied. The overall effect of the different herbicides was most pronounced in the Las Animas clay.

The effect of the rate of herbicide application when combined with the effect of soil type is shown in Fig. 2 and also in Table 6. Table 6 points out that the soils react similarly at all rates of herbicide application and nitrate nitrogen production increases (in general) with rate of herbicide application and with increase in the clay content of the soil.

The data arranged to show the relationship between soil type and sampling date are shown in Table 7. The data indicates that all soils react similarly to produce increasing amounts of nitrate nitrogen as the length of incubation of the soil sample increases.

The effect of sampling dates when combined with herbicides is given in Table 8. These data indicate that calcium cyanamide increased the amount of nitrate nitrogen in the sample at all sampling dates significantly

over 444E and Chloro-IPC. The latter two herbicides do not react similarly at all sampling periods, but the differences are small.

action between herbicide rate and kind of herbicide when the nontreated samples are compared with the first rate of all herbicide. Between the first and second and second and third rates the herbicides act similarly with CaCN2 producing the most nitrate nitrogen, and 444E producing slightly but not significantly less than Chloro-IPC.

The remaining undiscussed first order interactions may be examined in the tables in appendix II. The second order interactions are also given. In general the first order interactions show significant changes in the magnitude of nitrate nitrogen production but do not show any appreciable changes in effect. The second order interactions show a great deal of variation, some show no significance, some barely significant, and two are highly significant. In general, no additional information was obtained from the second order interactions. More data is necessary to interpret accurately these three factor interactions. As noted previously, the degrees of freedom for the third and higher order interactions were lumped and included in the error term so no attempt was made to explain these interactions.

Table 3 .-- TEMPERATURE BY HERBICIDE INTERACTION .

Herbicides	13°C ug N	25°C 03/1 g of	35°C soil	L.S.D. .05 level	L.S.D. .01 level
Check	9,82	12,59	15,76	0.66	1.09
CaCN2	14,06	19.66	26,74	0.66	1.09
CIPC	13.50	16.50	20,80	0.66	1,09
444E	12.74	16.06	21,82	0,66	1,09

Table 4. -- TEMPERATURE BY HERBICIDE RATE INTERACTION.

derbicide rate	13°C	25°C 3-n/1 g of	35°C soil	L.S.D.	L.S.D.
0	13.12	16.80	21.02	0.67	1.02
1	12.80	16.45	21.42	0.67	1.02
2	14.80	17.78	23.60	0.67	1.02
3	13.02	18.62	26.45	0.67	1.02

Table 5 .-- HERBICIDE BY SOIL TYPE INTERACTION .

Soil class	Check ug 1	CaCN2 NO3-N/1 g	CIPC of soi	444E 1	L.S.D. .05 level	L.S.D. .01 level
Sand	9.14	13.63	12.77	12.19	0.67	1.02
Loam	13.23	18.80	16.62	16.96	0.67	1.02
Clay	15.82	28.03	21.39	21.46	0.67	1.02

Table 6 .-- SOIL TYPE BY HERBICIDE RATE INTERACTION .

Herbicide rate	Sand ug NO	Loam 3/1 g of	Clay soil	L.S.D.	L.S.D.
0	12.18	17.65	21.12	0.67	1.02
1	12.45	15.87	22.35	0.67	1.02
2	13.60	18.58	24.00	0.67	1.02
3	13,25	17.78	27.07	0.67	1.02

Table 7. -- SAMPLING DATE BY SOIL TYPE INTERACTION.

Sampling date	Sand µg NO	Loam 3/1 g of	Clay soil	L.S.D. .05 level	L.S.D. .01 level
5 days	10.35	14.62	18.45	0.67	1.02
10 days	11.90	17.38	22.45	0.67	1.02
20 days	14.50	18.50	24.53	0.67	1.02
30 days	14.70	19.38	29,12	0.67	1.02

Table 8. -- HERBICIDE BY SAMPLING DATE INTERACTION.

	npling	CaCN2	CIPC	444E	L.S.D.	L.S.D.	
	iate	ug NO3/1 g of soil		.05 level	.01 level		
5	days	15.60	13.90	13.90	0.67	1.02	
10	days	19.06	16.98	15.68	0.67	1.02	
20	days	22.13	17.15	18.21	0.67	1.02	
30	days	23.81	19.68	19.68	0.67	1.02	

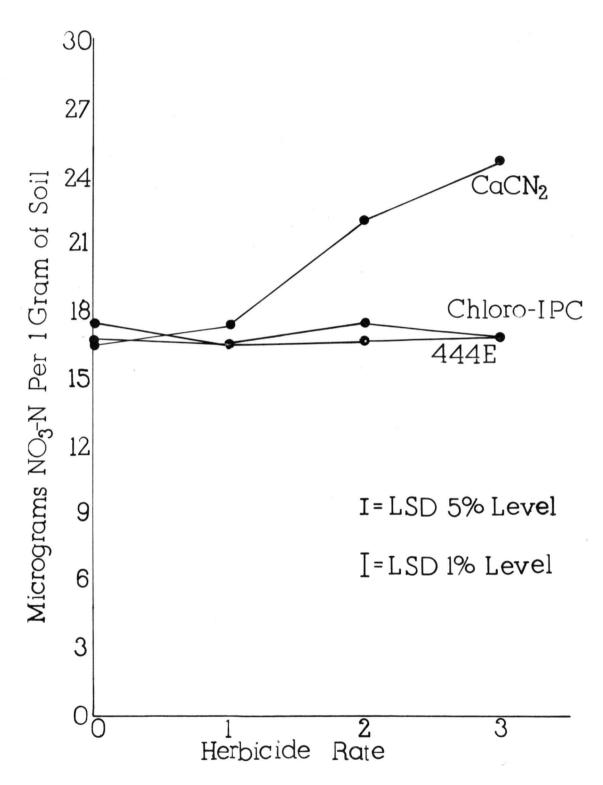


Fig. 1.--Herbicide by herbicide rate interaction.

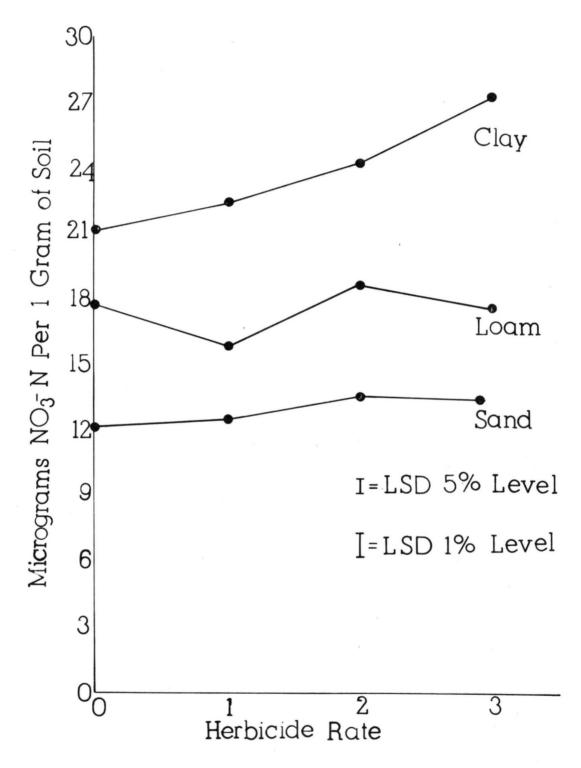


Fig. 2.--Soil type by herbicide rate interaction.

DISCUSSION OF RESULTS

Gilcrest sand, Rocky Ford loam, and Las Animas clay) and the nitrate nitrogen produced by the soil over an incubation period of 30 days and at controlled temperatures measured. There was no reduction of nitrate over that in the control soil, in fact a slight increase was noted in all cases. It was not the intent of this study to investigate the mechanism which altered the nitrate nitrogen content of soils. However, where possible, explanation of the mechanism is made.

found to contain a much higher level of nitrate than those soils to which the other herbicides had been added. It is reasonable that this should be so since CaCN2 contains 21.5 per cent nitrogen and is known to decompose and liberate nitrogen which eventually is changed to nitrate. A general breakdown of the decomposition of calcium cyanamide in the soil is given by Crowther and Richardson (7).

Galcium cyanamide which is present in the fertilizer in a crystalline form, dissolves with decomposition in water to give an acid salt and calcium

hydroxide:

In concentrated solutions a basic salt ultimately separates out in needles and free cyanamide is formed:

Free cyanamide may be regarded either H2N.CN or as carbodiimide HN C N H, and both forms are probably present in solution. It behaves as a weak monobasic acid; dibasic salts do not exist in solution.

An aqueous solution of pure cyanamide is relatively stable, but in the presence of acid or alkali or certain other catalysts it undergoes fairly rapid changes. In moderately alkaline solutions, especially when heated, it polymerises almost quantitatively to dicyanodiamide NH2

 ${\tt C}$ ${\tt NH}.$ The rate of reaction increases with the pH value ${\tt NH}.{\tt CN}$

up to about 9.6, but in still more alkaline conditions (pH greater than 10) the rate falls off rapidly, while hydrolysis to urea commences and becomes almost quantitative about pH value of 12.

In acid solutions free cyanamide is hydrolysed to urea, and the reaction is catalysed by many organic compounds, especially salts or oxides of iron or manganese. This is the formal change undergone in the soil.

Calcium cyanamide when applied to the soil usually decomposes into a form readily available for plant utilization. In the process it passes through a stage which is toxic to germinating weed seeds. In dry soils having a relatively high pH above 8.0, calcium cyanamide tends to polymerize into diciandiamid (H2NCN2)2. This compound has little value either as a fertilizer or herbicide because of its relative stability and insolubility.

For the other two herbicides, CIPC and 444E, the amount of nitrogen which may have been introduced by the application of the herbicides, was negligible.

There are a number of possible explanations for the increase in nitrate nitrogen over that in the control in those soils treated with CIPC and 444E. One explanation might be that partial sterilization of the system is brought about by the treatment. There are several theories to explain how partial sterilization increases soil fertility. The process may destroy the biological check which has maintained a given population at a certain level. The addition of a toxic material or other severe treatment serves to kill or severely restrict the activities of some members of the microbial population. It is known that the protozoa and some higher fungi are especially susceptible. The organisms which have been

killed serve as a high carbon-nitrogen ratio food source for the remaining portion of the population. Since these organisms are high in nitrogen, ammonia is released is nitrified and nitrate, in excess of that in a control soil, is found upon analysis. This may very well be the explanation for the increase in nitrate nitrogen found in this experimental study.

a function of both temperature and time. The temperatures used were well within the range normally found in the same soils in the field. Temperature increases normally would double the amount of nitrate produced for each 10 degree rise in temperature (degrees centigrade). The increases herein reported are somewhat less than this. The temperature range used is quite similar to that which would be encountered in the field from spring to late summer and the rate of nitrification coupled with the information available as to the effect of length of incubation gives a means of predicting, at least in the case of CaCN2, the amount of nitrate which would be available at a given length of time after application.

Each soil produces a characteristic and different curve for nitrate accumulation as a function of the variables studied in this experiment. The curve for a given soil is a function of many variables, in addition to those studied here. The immediate pretreatment of the sample prior to the initiation of the experiment is particularly important and was kept constant for all soils used. Field history, often difficult to obtain, would also condition the results which are obtained. The more clay in a soil, in general, the more favorable would the soil environment be for microorganisms. In general organic matter (food), water, and mineral nutrients increase with clay content while aeration may become less desirable.

When the effect of each herbicide was isolated and compared with the untreated soil, it was found that soils treated with CaCN2 had a significantly greater amount of nitrate nitrogen. Soils treated with CIPC and 444E were not statistically different from the control. However, in the following interactions: temperature by herbicide and soil type by herbicide, there was a significant increase in the amount of nitrate nitrogen when compared to untreated soil. It is probable that the interaction effect comes closer to measuring what actually takes place when herbicides are applied to the soil than isolating one factor and using it to represent the overall effect. Processes in nature seldom function as the result of the operation of a single variable. rather many variables are involved in a single measurement. In experimental research attempts are made to

eliminate or hold constant all variables except the one under consideration. Whether or not such idealized experimental conditions are actually produced is a matter for conjecture.

All herbicidal applications were made prior to incubation with one exception. A split application of CIPC was used in which the equivalent of three pounds per acre was applied prior to incubation, and then the same rate was applied to the same soil samples after they had incubated for one week. There was no indication of a reduction in the production of nitrate nitrogen resulting from this treatment. This would tend to indicate that repeated additions of the same type of herbicide (at rates commonly used for weed control) do not greatly influence microbial activity as measured by the specialized process of nitrate production. The practical recommendation would be that these herbicides could be used for a considerable number of years with little effect on the production of nitrate nitrogen and probably with little effect on the soil in general.

In summary it may be concluded that for the herbicides considered, there is no deleterious effect on the soil microbial population, particularly those organisms which produce nitrate nitrogen. The increase in nitrate nitrogen production would enhance crop growth and there is

little indication that there would be any residual effects from the frequent use of these materials. The study
indicates that if conditions were favorable, calcium
cyanamide could serve both as a herbicide and as a
fertilizer.

APPENDIX I

Coded Data

Bainbridge et.al. (4), states that in analyzing the results of a factorial experiment it is sometimes desirable to calculate separately a complete set of means and mean squares each relating to one degree of fredom. This complete analysis is most valuable in complex experimental designs which include confounding, different error components, and possibilities of error varying with factor level or replication, or where it is desirable to examine and allow for the effect of some uncontrolled variable. For example in the present experiment, the sum of squares associated with the temperature by herbicide interaction (containing four degrees of freedom) was 142.22. Using the tabular analysis described by Bainbridge et.al. (4), the sum of squares associated with each degree of freedom of this interaction are: 59.63 (linear by linear effect). 1.46 (linear by quadratic effect), 30.25 (quadratic by linear effect), and 0.88 (quadratic by quadratic effect), giving a more detailed study of the interaction. evident that two of the sum of squares of the temperature by herbicide interaction are not significant. Thus the degree of freedom associated with each of these two sums

of squares might very well be combined with the error term, which would tend to increase the precision of future experiments.

The generalized tabular method is illustrated in Table 9 for the factorial experiment used in this study. The coefficients used in each stage of the calculation are found in Tables 10 and 11. Referring now to Table 19, the raw data are given in the column headed NO3-N values. Column 1 was formed by first dividing the column headed by NO 3-N values into groups of threes to represent the three levels of factors (herbicides, soils, temperatures). The sums of each group of three numbers were obtained, and those numbers became the first 144 numbers of column 1. Those numbers were obtained by using coefficient 0 in Table 10. The second 144 numbers of column 1 were obtained by subtracting the first number (in the first group of three numbers) from the third or last number. Coefficient 1 in Table 10 was used in that step. The last 144 numbers in column 1 were obtained by subtracting two times the second number from the first number of the group of three numbers, and adding the third. Coefficient 2 of Table 10 was used in this step. This process was repeated over again for each group of three numbers in columns 1, 2, and 3. There are 144 groups containing three numbers in each of these columns.

Columns 4 and 5 were computed by dividing the columns into groups with four numbers each, as these represent factors which have four levels. The first 108 numbers in column 4 were obtained by summing the first four numbers of column 3. This is indicated as coefficient 0 Table 11, The second 108 numbers of column 4 were obtained by using coefficient 1 of Table 11 on the first four numbers of column 3. The third group of 108 numbers of column 4 were obtained by using coefficient 2 of Table 11 on the first four numbers of column 4 were obtained by using coefficient 3 of Table 11 on the first four numbers of column 3. The last 108 numbers of column 4 were obtained by applying coefficient 3 of Table 11 to the first four numbers of column 3. This process was repeated for all groups of four numbers in columns 3 and 4.

Table 10. -- ORTHOGONAL POLYNOMIAL COEFFICIENTS AND DIVISORS FOR 3-LEVEL FACTORS.

Coefficient number	1	3		Divisor
0	1	1	1	3
1	-1	0	1	2
3	1	-3	2	6

Table 11.--ORTHOGONAL POLYNOMIAL COEFFICIENT AND DIVISORS FOR 4-LEVEL FACTORS.

Coefficient					
numbe r	1	3	3	4	Divisor
0	1	1	1	1	4
1	-3	-1	1	3	30
2	1	-1	-1	1	4
3	-1	3	-3	1	30
Check	-2	2	-2	6	

The column headed divisor (Table 9) was calculated by using the divisors associated with the coefficients in Tables 10 and 11. Going back to the column headed RDHST (Table 9) and in the example below, it will be noted that the columns headed by R and D contain four factors. The divisors associated with the coefficients in Table 11 were used for the first two columns. The next three columns headed by the letters HST contain only three factors, thus the divisors associated with the coefficients from Table 10 were used on those three columns.

Type of	-	9%	**	~		Divisor of
treatment	K	D	H	۵.	T	each column Divisor
Number of						
factors in						
each treat						
ment	4	4	3	3	3	
Numerical code	0	0	0	0	0	4 4 3 3 3 432
for each	1	1	2	2	2	20 20 6 6 6 86,400
treatment	1	0	3	2	2	20 4 6 6 6 17,280
	0	0	0	1	1	4 4 3 2 2 192

The divisor associated with 0 treatment (when the treatment contains four factors) was found by looking for the number 0 in Table 11 under the heading coefficient number, and then reading across the table until the number four was found under the heading divisor. The divisor for any treatment containing four factors can be found in exactly the same manner. When the treatment contained only three factors the same method was used except Table 10 was used instead of Table 11. When the divisor of each column was found the product of these five numbers was the final divisor.

Each column was checked in the following manner in order to correct for mistakes. Beginning with the column headed NO3-N values, the sum was obtained. The same column was divided into groups containing three numbers as done before. The first number of all groups (there are 144 groups hence 144 numbers) were added together. All second numbers of each group and all third numbers of each group were summed in like manner. This gave three sums which were written at the end of this column and labeled subtotal. The sum of these three sums was obtained, as shown at the end of Table 9. In Table 10, in the row prefixed by the word "check", the numbers 1, -1, and 3 were multiplied by the subtotal sums of the column headed NO2-N values in such a way that the first

subtotal sum was multiplied by 1, the second by -1 and the third by 3. The following example was taken from Table 9. The first, second, and third subtotal sums were 1,209, 1,567, and 2,081 respectively. The multiplier values taken from Table 10 were 1, -1 and 3. (1209 x 1 plus 1567 x -1 plus 2081 x 3 equaled 5885). When the sum of the subtotal sums for column 1 was compared to the number 5,885, they were found to be the same. A mistake would be present if the two numbers were not identical. This process was the same for all of the columns except 4 and 5. For these columns the subtotal sums were multiplied by the numbers in the row prefixed by the word "check" in Table 11.

fine sum of squares associated with each single degree of freedom was obtained by squaring each item in column 5 and dividing by the divisor associated with the particular treatment.

Compared with the more orthodox method of calculating the sums of squares from a number of two-way,
three-way, four-way, and even five-way tables the tabular
method presents much less opportunity for copying errors,
and also possesses the advantage that it is completely
self-checking. Main effects and interactions can also be
obtained from column 5. In this study these were obtained
from 2 and 3-way tables.

Table 9 .-- ANALYSIS OF VARIANCE USING ORTHOGONAL POLYNOMIALS (4).

							- December 1980) - 44		
RDHST	Treatment	N03-N	I	II	III	IV	V	Divisor	Sum of
-		values	-						squares
			4 49					432	F1 (07 F0
00000		2	17 32	81 69	236 273	1146 1140	4857 872	288	54,607.52 2,640.22
00002		6	32	86	308	1264	156	864	28.17
00010		7	15	91	329	1307	969	288	3,260.28
00011		13	28	95	223	178	342	192	609.19
00012		12	26	87	273	194	150	576	39.06
00020	And the state of t	10	19	101	301	198	141	864	23.01
00021		10	31	101	343	302	278	576	134.17
00022		12	36	106	273	12	30	1728	0.52
00100		4	23	101	293	30	-295	288	302.17
00101	HITL	5	33	114	340	64	-107	192	59.63
00102	H1T2	6	35	114	358	50	29	576	1.46
00110	H1S1	8	24	76	245	201	-154	192	123.52
00111	H1S1T1	12	32	77	325	223	- 80	128	50.00
00112	H1S1T2	8	39	70	345	234	- 58	384	8.76
00120		4	21	91	392	311	-130	576	29.34
1.4	H1S2T1	11	30	91	36		-116	384	35.04
	H1S2T2	11	36	91	28	74	- 58	1152	2.92
00200		4	26	106	44	60	285	864	94.01
00201	April 100 mm. The	7	35	93	70	122	215	576	80.25
00202		8	40	102	40	12	39	1728	00.88
00210		8	25	120	32	64	192	576	64.00
	H2S1T1	10	35	113	35	60	36	384	3.38
	H2S1T2	13	41	110	87	14	-114	1152	11.28
00220		9	26	107	36	- 45	60	1728	2.08
	H2S2T1	11	37	88	53	69	8	1152	0.06
	H2S2T2	16	43	78	52	10	30	3456	0.26
01000	Contract of the Contract of th	9	27	105	57	107	1465	2160	993.62
01001		6	27	101	39	40	542	1440	204.00
01010		8	47 27	87 140	80 67	74 36	130 415	4320 1440	3.91 119.60
	DISITI	10	37	102	116	128	292	960	88.82
	DISIT2	14	50	98	- 28		- 32	2880	0.36
01020		9	24	142	36	- 24	427	4320	42.21
	D1S2T1	12	40	103	- 4	- 56	368	2889	47.02
	D1S2T2	14	50	113	8		- 44	8640	0.22
011001	Chief Staff Staff State	12	19	87	- 14		-177	1440	21.76
	DIHITI	4	25	79	12		- 73	960	5.55
	D1H1T2	8	32	79	7	-118	63	2880	1.38
	D1H1S1	10	18	142	25	-176		960	0.34
	DIHISITI	10	26	95	6	12	8	640	0.10
	D1H1S1T2	12	33	88	47		- 54	1920	1.52
01120	D1H1S2	9	20	151	- 8		-198	2880	13.61
	D1H1S2T1	12	20	90	19	-128	44	1920	1.01
01122	D1H1S2T2	18	30	104	11		-174	5760	5.26

Table 9 .-- ANALYSIS OF VARIANCE USING ORTHOGONAL POLYNOMIALS (4).-- Continued.

RDHST	Treatment	NO3-N values		II	III	IV	V	Divisor	Sum of squares
01200	2010	6	23	173	34	- 20	283	4320	18.54
	D1H2T1	6	27	113	- 9	- 7	65	2880	1.47
	D1H2T2	9	41	106	14	40	145	8640	2.43
	D1H2S1	9	20	11	43	14	112	2880	4.36
	D1H2S1T1	9	29	9	42	-	76	1920	3.01
-	D1H2S1T2		42		47		-254	5760	11.20
		12		16	69				
	D1H2S2	10	24	9		-128 4	-296	8640	10.14
	D1H2S2T1	10	29	7	38	-	-136	5760	3.21
	D1H2S2T2	16	38	12	54	- 7	106	17280	0.65
02000	The second secon	8	29	15	52	- 12	- 59	432	8.06
02001		8	32	14	79	~ 65	90	288	28.13
02002		10	45	15	51	- 10		864	6.34
02010	26. 27. 40. 14	9	25	19	56	- 47	43	288	6.42
- CAL III - CAL III	D2S1T1	14	27	28	49	- 26	18	192	1.69
	D2S1T2	12	41	23	78	25	166	576	47.84
02020		9	24	13	50	- 14	67	864	5.20
	D2S2T1	12	34	8	85	- 5	18	576	0.56
	D2S2T2	19	44	19	78	- 7	- 2	1728	0.00
02100		8	26	8	98	-104	33	288	3.78
	D2H1T1	9	39	11	6	6	25	192	3.26
	P2H1T2	8	55	13	22	- 21	9	576	0.14
	D2H1S1	9	28	11	26	- 18	6	192	0.19
	D2H1S1T1	12	34	8	32	- 83	20	128	3.13
	D2H1S1T2	14	51	16	5	- 8	- 90	384	21.09
	D2H1S2	10	24	36	14	- 5	18	576	0.56
	D2H1S2T1	12	35	25	19	32	- 20	384	1.04
	D2H1S2T2	19	51	26	36	- 77	- 6	1152	0.03
02200		8	24	13	9	9	- 23	864	0.61
	D2H2T1	9	35	6	15	18	15	576	0.39
02202	D2H2T2	9	48	17	16	82	7	1728	0.03
02210	D2H2S1	10	23	22	20	176	4	576	0.03
02211	D2H2S1T1	13	29	19	23	4	0	384	0.00
02212	D2H2S1T2	14	36	12	25	38	46	1152	1.84
02220	D2H2S2	9	19	21	25	51	4	1728	0.01
02221	D2H2S2T1	15	26	13	49	122	- 48	1152	2.00
02222	D2H2S2T2	19	33	18	10	- 36	34	3456	0.33
03000	D3	7	22	17	- 12	24	55	2160	1.40
03001	D3T1	10	41	11	8	25	164	1440	18.68
03002		10	42	29	6	26	520	4320	62.59
03010		8	20	26	23	6	175	1440	21.27
	D3S1T1	9	42	9	0	- 5	64	960	4.27
	D3S1T2	10	39	4	19	45	-104	2880	3.76
03020		9	20	48	22		- 41	4320	0.39
	D3S2T1	15	30	19	27		-124	2880	5.34
	D3S2T2	23	37	13	5		-188	8640	4.09
	The second secon	-		-	-				

Table 19 .-- ANALYSIS OF VARIANCE USING ORTHOGONAL POLYNOMIALS (4).-- Continued.

								Control of the Contro	
RDHST	Treatment	N03-N	I	II	III	IV	V	Divisor	Sum of squares
-		values		process of the same		-			2400200
03100	B2011	6	35	46	- 2	- 30	- 19	1440	0.25
	D3H1T1	9	48	6	30	77	- 51	960	2.71
	D3H1T2	12	57	15	29	24	- 19	2880	0.13
	D3H1S1	10	29	65	- 11	7	- 86	960	7.70
	D3H1S1T1	10	33	26	- 15	-	-104	640	16.90
	D3H1S1T2	17	40	25	11	- 97	- 38	1920	0.75
	D3H1S2	10	24 -		- 37	8	- 46	2880	0.73
	D3H1S2T1	15	34 -	-	- 12	- 9	168	1920	14.70
	D3H1S2T2	25	40	2	- 13	- 11	62	5760	0.67
03200		7	33	7	17	74	-299	4320	20.69
	D3H2T1	7	39	17	10	- 8	-145	2880	7.30
	D3H2T2	10	70	12	18	- 19	-185	8640	3.96
	D3H2S1	10	29 -		22	- 12	4	2880	0.01
-	D3H2S1T1	15	25	2	19	47	52	1920	1.41
	D3H2S1T2	15	49 -		3	6	202	5760	7.08
	D3H2S2	9	28 -	-	- 46	- 39	208	8640	5.01
	D3H2S2T1	17	36	12	- 5	- 14	188	5760	6.14
	D3H2S2T2	24	49 -	-	58	77	142	17280	1.17
10000	The company of the contract of	3	17 -		14	314	607	2160	170.58
10001		10	24	2	31	388	376	1440	98.18
10002		6	46 -	-	42	302	148	4320	5.07
10010	200 A DEC 200 - 200	6	20	10	20	461	341	1440	80.75
	RISITI	9	28	1	6	118	94	960	9.20
	R1S1T2	10	31	1	- 2	144	2	2880	0.00
10020		8	22	13	8	62	397	4320	36.48
	R1S2T1	10	25 -	6	28	218	226	2880	17.73
	R1S2T2	14	32	0	1	68	- 26	8640	0.08
10100	Marie Control of Control of Control	4	23	12	26	112	-683	1440	323.95
10101	R1H1T1	7	34	5	35	- 16	-423	960	186.38
	R1H1T2	7	85	8	12	- 34	85	2880	2.51
	R1H1S1	8	27	11	- 3	83	-456	960	216.60
10111	R1H1S1T1	9	32	10	- 13	121	-212	640	70.23
10112	R1H1S1T2	. 9	36 -	15	10	74	126	1920	8.27
10120	R1H1S2	10	21	18	42	137	-272	2880	25.69
10121	R1H1S2T1	9	32	17	21	82	-264	1920	36.30
10122	R1H1S2T2	14	35	12	17	98	-170	5760	5.02
10200	R1H2	4	32 -	13	49	34	565	4320	73.89
10201	R1H2T1	7	41 -	3	41	78	367	2880	46.77
	R1H2T2	9	78	8	14	8	187	8640	4.05
	R1H2S1	2	25	13	12	16	470	2880	76.70
	R1H2S1T1	8	25 -	-	26	2	256	1920	34.13
	R1H2S1T2	10	40	17	2	- 58	-418	5760	30.33
	R1H2S2	. 7	26	6	7	161	202	8640	4.72
	R1H2S2T1	10	35 •	_	0	31	172	5760	5.14
10222	R1H2S2T2	13	43	10	7	206	238	17280	3.28

Table 9 .-- ANALYSIS OF VARIANCE USING ORTHOGONAL POLYNOMIALS (4) .-- Continued.

RDHST	Treatment	NO3-N values	I	11	III	IA	v	Divisor	Sum of squares
11000	R1D1	7	33	4	- 38	29	355	10800	11.67
40.00	R1D1T1	6	52	17	- 21	76	218	7200	6.60
	R1D1T2	10	88	13	- 31	42	-434	21600	8.72
	RID1S1	11	28	- 2	- 20	158	115	7200	1.84
	RIDISITI	9	37	- 12	16	92	- 76	4800	1.20
	RID1S1T2	7	48	5	23	- 22	-212	14400	3.12
	R1D1S2	10	24	- 7	1	-128	-221	21600	2.26
	RID1S2T1	12	35	8	21	122	164	14400	1.87
11022	RID1S2T2	19	47	13	11	- 16	268	43200	1.66
	R1D1H1	6	4	15	5		-617	7200	52.87
11101	RIDIHIT1	7	5	11	- 4	- 16	- 53	4800	0.59
11102	RID1H1T2	7	2	17	5	- 24	487	14400	16.47
	R1D1H1S1	8	2	12	13	-170	-128	4800	3.41
11111	RID1H1S1T1	9	0	15	- 6	- 6	40	3200	0.50
	RID1H1S1T2	12	7	15	0	- 48	166	9600	2.87
11120	R1D1H1S2	11	4	14	- 4	31	292	14400	5.92
11121	RID1H1S2T1	14	5	16	- 10	- 50	124	9600	1.60
11122	RID1H1S2T2	17	7	1.7	- 29	- 66	-458	28800	7.28
11200	R1D1H2	6 -	1	20	- 18	- 34	943	21600	41.17
11201	RID1H2T1	8	5	23	- 42	117	569	14400	22.48
11202	R1D1H2T2	10	5	26	- 29	46	325	43200	2.45
11210	R1D1H2S1	9 -	4	13	- 8	12	166	14400	1.91
11211	RID1H2S1T1	9	2	15	- 54	11	-292	9600	8.88
11212	R1D1H2S1T2	11	9	10	- 47	-21	-26	28800	0.02
11220	R1D1H2S2	9	3	18	- 67	- 20	-290	43200	1.95
11221	R1D1H2S2T1	13	3	22	5	- 8	-592	28800	12.17
11222	R1D1H2S2T2	16	6	14	3	5	-386	86400	1.72
12000	R1D2	8	2	16	0	6	- 45	2160	0.94
12001	R1D2T1	11	3	16	4	5	-150	1440	15.63
12002	R1D2T2	10	10	20	6	- 18	150	4320	5.21
12010	R1D2S1	12	0	29	5	- 31	-101	1440	7.08
12011	R1D2S1T1	9	5	23	5	- 48	86	960	7.70
12012	R1D2S1T2	11	9	27	- 10	43	166	2830	9.57
12020	R1D2S2	12	1	24	4	-108	- 9	4320	0.02
12021	R1D2S2T1	11	4	13	- 10	- 25	- 6	2880	0.01
12022	R1D2S2T2	22	10	14	- 3	- 69	186	8640	4.00
12100	R1D2H1	6	3	20	12	4	41	1440	1.17
12101	R1D2H1T1	9	2	1.9	~ 22	- 42	37	960	1.43
12102	R1D2H1T2	10	14	17	- 35	51	- 79	2880	2.17
12110	R1D2H1S1	10	6	22	- 31	28	60	960	3.75
12111	R1D2H1S1T1	11	7	11	- 40	7	- 16	640	0.40
12112	R1D2H1S1T2	6	15	16	17	36	- 18	1920	0.17
12120	R1D2H1S2	10	3	37	5	- 25	152	2880	8.02
12121	R1D2H1S2T1	13	5	20	- 4	- 90	- 80	1920	3.33
19199	R1D2H1S2T2	18	15	21	- 2	~ 95	-238	5760	9.83
	R1D2H2		3		6	-109			

Table 9 .-- ANALYSIS OF VARIANCE USING ORTHOGONAL POLYNOMIALS (4).-- Continued.

RDHST Treatment								**************************************		
12201 RID2H2T1	DDUCT	Trastmant	NO2-N	T	TT	TTT	TV	17	Divisor	Sum of
12201 RID2H2T1	MUNGE	er a a mineri		46	**		TA	v	PTATOOT	
12202 RIDZHZT2	-		varues	-	-	-				adoarea
12202 RIDZHZT2	12201	R1D2H2T1	8	4	11	- 9	58	- 9	2880	0.03
12210 RID2H2SI		and the same of th				-				
12211 RID2H2SIT1										
12212 RID2H2S1T2	The little state of the state of	Season like discounting on the								
12220 RID2H2S2		the second second second second						-		
12221 RID2H2S2T1		The second secon			-					
12222 RID2H2S2T2										
13000 RID3		The state of the s								
13001 RID3T1										
13002 RID3T2										
13010 R1D3S1										
13011 RID3S1T1 10		The state of the s								
13012 RID3S1T2		The state of the s								
13020 RID3S2										
13021 RID3S2T1										
13022 RID3S2T2		A					-			
13100 RID3H1										
13101 RlD3HlT1	The state of the s									
13102 R1D3H1T2		The second secon					-			
13110 R1D3H1S1		the state of the state of the state of								
13111 R1D3H1S1T1 10 - 4 9 - 6 - 10 - 200 3200 12.50 13112 R1D3H1S1T2 15 8 12 - 16 - 139 - 418 9600 18.20 13120 R1D3H1S2 10 2 3 - 19 8 - 116 14400 0.93 13121 R1D3H1S2T1 16 5 1 - 48 - 101 48 9600 0.24 13122 R1D3H1S2T2 25 9 1 - 29 - 121 - 206 28800 1.47 13200 R1D3H2 6 5 6 - 32 - 82 61 21600 0.17 13201 R1D3H2T1 8 13 5 5 100 - 137 14400 1.30 13202 R1D3H2T2 10 18 3 - 3 - 39 - 245 43200 1.39 13210 R1D3H2S1 10 4 8 1 - 130 322 14400 7.20 13211 R1D3H2S1T1 9 6 4 1 - 67 436 9600 19.89 13212 R1D3H2S1T2 16 15 7 - 2 158 358 28800 4.45 13220 R1D3H2S2 9 4 13 - 3 - 71 430 43200 4.28 13221 R1D3H2S2T1 17 6 11 - 1 20 76 28800 0.20 13222 R1D3H2S2T2 25 16 12 - 1 - 1 178 86400 0.37 20000 R2 4 6 0 - 1 - 16 49 432 5.56 20001 R2T1 10 1 10 - 6 - 8 88 288 26.89 20002 R2T2 10 6 - 1 - 6 - 2 - 32 864 1.19 20011 R2S1T1 8 2 7 - 13 34 74 192 28.52 20012 R2S2T1 14 7 1 - 2		The second second second second			1.000					
13112 R1D3H1S1T2 15 8 12 - 16 -139 -418 9600 18.20 13120 R1D3H1S2 10 2 3 - 19 8 -116 14400 0.93 13121 R1D3H1S2T1 16 5 1 - 48 -101 48 9600 0.24 13122 R1D3H1S2T2 25 9 1 - 29 -121 -206 28800 1.47 13200 R1D3H2 6 5 6 - 32 - 82 61 21600 0.17 13201 R1D3H2T1 8 13 5 5 100 -137 14400 1.30 13210 R1D3H2S1 10 4 8 1 -130 322 14400 7.20 13211 R1D3H2S1T1 9 6 4 1 - 67 436 9600 19.89 13212 R1D3H2S2T1 16 15 7 - 2 158 358 28800 4.45 13220 R1D3H2S2T1 17 6 11 - 1 20 76 28800 0.20 13222 R1D3H2		the second secon								
13120 R1D3H1S2								-		
13121 R1D3H1S2T1 16 5 1 - 48 -101 48 9600 0.24 13122 R1D3H1S2T2 25 9 1 - 29 -121 -206 28800 1.47 13200 R1D3H2 6 5 6 - 32 - 82 61 21600 0.17 13201 R1D3H2T1 8 13 5 5 100 -137 14400 1.30 13202 R1D3H2T2 10 18 3 - 3 - 39 -245 43200 1.39 13210 R1D3H2S1 10 4 8 1 -130 322 14400 7.20 13211 R1D3H2S1T1 9 6 4 1 - 67 436 9600 19.89 13212 R1D3H2S1T2 16 15 7 - 2 158 358 28800 4.45 13220 R1D3H2S2 9 4 13 - 3 - 71 430 43200 4.28 13221 R1D3H2S2T1 17 6 11 - 1 20 76 28800 0.20 13222 R1D3H2S2T2 25 16 12 - 1 - 1 178 86400 0.37 20000 R2 4 6 0 - 1 - 16 49 432 5.56 20001 R2T1 10 1 10 - 6 - 8 88 288 26.89 20002 R2T2 10 6 - 1 - 6 - 2 -32 864 1.19 20010 R2S1 13 - 3 7 1 - 33 55 288 10.50 20011 R2S1T1 8 2 7 - 13 34 74 192 28.52 20012 R2S2T1 14 7 1 - 29 60 - 98 576 16.67 20020 R2S2 14 4 8 - 3 - 12 - 17 864 0.33 20021 R2S2T1 14 10 6 - 20 8 58 576 5.84 20022 R2S2T1 14 10 6 - 20 8 58 576 5.84		The state of the s								
13122 R1D3H1S2T2										
13200 R1D3H2 6 5 6 - 32 - 82 61 21600 0.17 13201 R1D3H2T1 8 13 5 5 100 -137 14400 1.30 13202 R1D3H2T2 10 18 3 - 3 - 39 -245 43200 1.39 13210 R1D3H2S1 10 4 8 1 -130 322 14400 7.20 13211 R1D3H2S1T1 9 6 4 1 -67 436 9600 19.89 13212 R1D3H2S2T2 16 15 7 -2 158 358 28800 4.45 13220 R1D3H2S2T1 17 6 11 -1 20 76 28800 0.20 13222 R1D3H2S2T2 25 16 12 -1 -1 178 86400 0.37 20000 R2 4 6 0 -1 -16 49 432 5.56 20001 R2T1 10 6 -1 -6 -8 88 288 26.89 20010 R2S1 13 <										
13201 RlD3H2T1 8 13 5 5 100 -137 14400 1.30 13202 RlD3H2T2 10 18 3 -3 -39 -245 43200 1.39 13210 RlD3H2S1 10 4 8 1 -130 322 14400 7.20 13211 RlD3H2S1T1 9 6 4 1 -67 436 9600 19.89 13212 RlD3H2S1T2 16 15 7 -2 158 358 28800 4.45 13220 RlD3H2S2 9 4 13 -3 -71 430 43200 4.28 13221 RlD3H2S2T1 17 6 11 -1 20 76 28800 0.20 13222 RlD3H2S2T2 25 16 12 -1 -1 178 86400 0.37 20000 R2 4 6 0 -1 -16 49 432 5.56 20001 R2T1 10 1 10 -6 -8 88 288 26.89 20012 R2S1 13 -3 7 -1 -33 55 288 10.50 20011 R2S1T1 8 2 7 -13 34 74 192 28.52		The second secon								
13202 RlD3H2T2 10 18 3 - 3 - 39 -245 43200 1.39 13210 RlD3H2S1 10 4 8 1 -130 322 14400 7.20 13211 RlD3H2S1T1 9 6 4 1 - 67 436 9600 19.89 13212 RlD3H2S1T2 16 15 7 - 2 158 358 28800 4.45 13220 RlD3H2S2 9 4 13 - 3 - 71 430 43200 4.28 13221 RlD3H2S2T1 17 6 11 - 1 20 76 28800 0.20 13222 RlD3H2S2T2 25 16 12 - 1 - 1 178 86400 0.37 20000 R2 4 6 0 - 1 - 16 49 432 5.56 20001 R2T1 10 1 10 - 6 - 8 88 288 26.89 20010 R2S1 13 - 3 7 1 - 33 55 288 10.50 20011 R2S1T1 8 <td>13201</td> <td>R1D3H2T1</td> <td></td> <td></td> <td></td> <td></td> <td></td> <td></td> <td></td> <td></td>	13201	R1D3H2T1								
13210 R1D3H2S1 10 4 8 1 -130 322 14400 7.20 13211 R1D3H2S1T1 9 6 4 1 -67 436 9600 19.89 13212 R1D3H2S1T2 16 15 7 -2 158 358 28800 4.45 13220 R1D3H2S2 9 4 13 -3 -71 430 43200 4.28 13221 R1D3H2S2T1 17 6 11 -1 20 76 28800 0.20 13222 R1D3H2S2T2 25 16 12 -1 -1 178 86400 0.37 20000 R2 4 6 0 -1 -16 49 432 5.56 20001 R2T1 10 1 10 -6 -8 88 288 26.89 20010 R2S1 13 -3 7 1 -33 55 288 10.50 20011 R2S1T1 8 2 7 -13 34 74 192 28.52 20012 R2S1T2 14 7	13202	R1D3H2T2	10							
13211 R1D3H2S1T1 9 6 4 1 -67 436 9600 19.89 13212 R1D3H2S1T2 16 15 7 -2 158 358 28800 4.45 13220 R1D3H2S2 9 4 13 -3 -71 430 43200 4.28 13221 R1D3H2S2T1 17 6 11 -1 20 76 28800 0.20 13222 R1D3H2S2T2 25 16 12 -1 -1 178 86400 0.37 20000 R2 4 6 0 -1 -16 49 432 5.56 20001 R2T1 10 1 10 -6 -8 88 288 26.89 20002 R2T2 10 6 -1 -6 -2 -32 864 1.19 20010 R2S1 13 -3 7 1 -33 55 288 10.50 20011 R2S1T1 8 2 7 -13 34 74 192 28.52 20012 R2S2T1 14 7 <t< td=""><td>13210</td><td>R1D3H2S1</td><td>10</td><td>4</td><td></td><td>1</td><td>-130</td><td>322</td><td></td><td></td></t<>	13210	R1D3H2S1	10	4		1	-130	322		
13220 R1D3H2S2 9 4 13 - 3 - 71 430 43200 4.28 13221 R1D3H2S2T1 17 6 11 - 1 20 76 28800 0.20 13222 R1D3H2S2T2 25 16 12 - 1 - 1 178 86400 0.37 20000 R2 4 6 0 - 1 - 16 49 432 5.56 20001 R2T1 10 1 10 - 6 - 8 88 288 26.89 20010 R2S1 13 - 3 7 1 - 33 55 288 10.50 20011 R2S1T1 8 2 7 - 13 34 74 192 28.52 20012 R2S1T2 14 7 1 - 29 60 - 98 576 16.67 20020 R2S2 14 4 8 - 3 - 12 - 17 864 0.33	13211	R1D3H2S1T1	9	6	4	1	- 67	436	9600	
13221 R1D3H2S2T1 17 6 11 - 1 20 76 28800 0.20 13222 R1D3H2S2T2 25 16 12 - 1 178 86400 0.37 20000 R2 4 6 0 - 1 - 16 49 432 5.56 20001 R2T1 10 1 10 - 6 - 8 88 288 26.89 20002 R2T2 10 6 - 1 - 6 - 2 -32 864 1.19 20010 R2S1 13 - 3 7 1 - 33 55 288 10.50 20011 R2S1T1 8 2 7 - 13 34 74 192 28.52 20012 R2S1T2 14 7 1 - 29 60 - 98 576 16.67 20020 R2S2 14 4 8 - 3 - 12 - 17 864 0.33 20021 R2S2T1	13212	R1D3H2S1T2	16	15	7	- 2	158	358	28800	4.45
13221 R1D3H2S2T1 17 6 11 - 1 20 76 28800 0 20 13222 R1D3H2S2T2 25 16 12 - 1 - 1 178 86400 0 37 20000 R2 4 6 0 - 1 - 16 49 432 5.56 20001 R2T1 10 1 10 - 6 - 8 88 288 26.89 20002 R2T2 10 6 - 1 - 6 - 2 -32 864 1.19 20010 R2S1 13 - 3 7 1 - 33 55 288 10.50 20011 R2S1T1 8 2 7 - 13 34 74 192 28.52 20012 R2S1T2 14 7 1 - 29 60 - 98 576 16.67 20021 R2S2T1 14 4 8 - 3 - 12 - 17 864	13220	R1D3H2S2	9	4	13	- 3			43200	
13222 R1D3H2S2T2 25 16 12 - 1 - 1 178 86400 0.37 20000 R2 4 6 0 - 1 - 16 49 432 5.56 20001 R2T1 10 1 10 - 6 - 8 88 288 26.89 20002 R2T2 10 6 - 1 - 6 - 2 -32 864 1.19 20010 R2S1 13 - 3 7 1 - 33 55 288 10.50 20011 R2S1T1 8 2 7 - 13 34 74 192 28.52 20012 R2S1T2 14 7 1 - 29 60 - 98 576 16.67 20020 R2S2 14 4 8 - 3 - 12 - 17 864 0.33 20021 R2S2T1 14 10 6 - 20 8 58 576 5.84	13221	R1D3H2S2T1	17	6						
20000 R2 4 6 0 - 1 - 16 49 432 5.56 20001 R2T1 10 1 10 - 6 - 8 88 288 26.89 20002 R2T2 10 6 - 1 - 6 - 2 -32 864 1.19 20010 R2S1 13 - 3 7 1 - 33 55 288 10.50 20011 R2S1T1 8 2 7 - 13 34 74 192 28.52 20012 R2S1T2 14 7 1 - 29 60 - 98 576 16.67 20020 R2S2 14 4 8 - 3 - 12 - 17 864 0.33 20021 R2S2T1 14 10 6 - 20 8 58 576 5.84 20022 R2S2T2 20 3 2 - 7 - 52 190 1728 20.89 </td <td>13222</td> <td>R1D3H2S2T2</td> <td>25</td> <td>16</td> <td>12</td> <td></td> <td></td> <td>178</td> <td>86400</td> <td></td>	13222	R1D3H2S2T2	25	16	12			178	86400	
20001 R2T1 10 1 10 - 6 - 8 88 288 26.89 20002 R2T2 10 6 - 1 - 6 - 2 -32 864 1.19 20010 R2S1 13 - 3 7 1 - 33 55 288 10.50 20011 R2S1T1 8 2 7 - 13 34 74 192 28.52 20012 R2S1T2 14 7 1 - 29 60 - 98 576 16.67 20020 R2S2 14 4 8 - 3 - 12 - 17 864 0.33 20021 R2S2T1 14 10 6 - 20 8 58 576 5.84 20022 R2S2T2 20 3 2 - 7 - 52 190 1728 20.89	20000	R2					- 16		432	
20002 R2T2 10 6 - 1 - 6 - 2 -32 864 1.19 20010 R2S1 13 - 3 7 1 - 33 55 288 10.50 20011 R2S1T1 8 2 7 - 13 34 74 192 28.52 20012 R2S1T2 14 7 1 - 29 60 - 98 576 16.67 20020 R2S2 14 4 8 - 3 - 12 - 17 864 0.33 20021 R2S2T1 14 10 6 - 20 8 58 576 5.84 20022 R2S2T2 20 3 2 - 7 - 52 190 1728 20.89	20001	R2T1	10							
20010 R2S1 13 - 3 7 1 - 33 55 288 10.50 20011 R2S1T1 8 2 7 - 13 34 74 192 28.52 20012 R2S1T2 14 7 1 - 29 60 - 98 576 16.67 20020 R2S2 14 4 8 - 3 - 12 - 17 864 0.33 20021 R2S2T1 14 10 6 - 20 8 58 576 5.84 20022 R2S2T2 20 3 2 - 7 - 52 190 1728 20.89	20002	R2T2	10		1		- 2	-32		
20011 R2S1T1 8 2 7 - 13 34 74 192 28.52 20012 R2S1T2 14 7 1 - 29 60 - 98 576 16.67 20020 R2S2 14 4 8 - 3 - 12 - 17 864 0.33 20021 R2S2T1 14 10 6 - 20 8 58 576 5.84 20022 R2S2T2 20 3 2 - 7 - 52 190 1728 20.89			13 -		7	1	- 33	55	288	
20012 R2S1T2 14 7 1 - 29 60 - 98 576 16.67 20020 R2S2 14 4 8 - 3 - 12 - 17 864 0.33 20021 R2S2T1 14 10 6 - 20 8 58 576 5.84 20022 R2S2T2 20 3 2 - 7 - 52 190 1728 20.89	20011	R2S1T1		2	7	- 13	34			
20020 R2S2 14 4 8 - 3 - 12 - 17 864 0.33 20021 R2S2T1 14 10 6 - 20 8 58 576 5.84 20022 R2S2T2 20 3 2 - 7 - 52 190 1728 20.89	20012	R2S1T2	14		1	- 29	60	- 98		
20021 R2S2T1 14 10 6 - 20 8 58 576 5.84 20022 R2S2T2 20 3 2 - 7 - 52 190 1728 20.89	20020	R2S2	14	4	8	- 3	- 12	- 17		
20022 R2S2T2 20 3 2 - 7 - 52 190 1728 20.89	20021	R2S2T1	14	10		- 20				
			20			- 7	- 52		1728	20.89
	20100	R2H1	9	2	6	9	- 8	- 19	288	1.25

Table 9 .-- ANALYSIS OF VARIANCE USING ORTHOGONAL POLYNOMIALS (4).-- Continued.

RDHST	Treatment	NO3-N values	I	11	III	IV	V	Divisor	Sum of squares
20101	R2H1T1	0	2.3	7		- 16	-125	192	01 20
	R2H1T1	8	11	7	- 3 - 9	- 14	83		81.38 11.96
		9	9	18				576	28.52
	R2H1S1 R2H1S1T1	9						192	
	R2H1S1T1	-	10	0 5	- 1	11 24	- 42	128	13.78
	R2H1S112	11	8	29	- 17 - 17		88 -106	384 576	20.17 19.51
	R2H1S2T1	9	5	4	- 7	- 15 - 10			
	R2H1S2T2	17	1 -	0	-		- 38 -112	384	3.76 10.89
20122	and the same of th		6		0-	- 2	85	1152	
	R2H2T1	3	5	10	6		37	864	8.36
	R2H2T1	9	3	8	- 25 - 1	22 20	-	576 1728	2.38 2.01
	R2H2S1	3	13		- 1	26	112	576	
	R2H2S1T1		1	29			74		21.78 14.26
	R2H2S1T1	10	5	11	25	54		384	1.000
		13	7	9	4	66	- 32 100	1152	0.89
	R2H2S2	9	5	12	8	. 5		1728 1152	5.79
	R2H2S2T1 R2H2S2T2	12	6 -	7	5	- 11	70		4.25
	and the same and the same and the same	12	7	5	- 1	112	136	3456	5.35
21000	The second secon	7	4 -	-	- 26	- 39	85	2160	3.34
	R2D1T1	6	3 -	9	9	28	130	1440	11.74
and later or the same	R2D1T2	9	10	3	- 6	- 48	- 62	4320	0.89
	R2D1S1	10	5	2	- 10	42	25	1440	0.43
	R2D1S1T1	10 -	6	7	2	- 4	28	960	0.82
	R2D1S1T2	21	12 -	1	- 2	- 22	- 68	2880	1.61
	R2D1S2 R2D1S2T1	10	7	5	15	- 38	- 47	4320	0.51
		13	8	5	0	46	- 32	2880	0.36
	R2D1S2T2	19	14 -		- 20	12	- 32	8640	0.12
AC 100 11 11 11	R2D1H1	6	4	13	- 11	17	- 97	1440	6.53
	R2D1H1T1	7	G	9	- 48	- 12	- 39	960	1.58
	R2D1H1T2		22	1	- 29	2	-103	2880	3.68
	R2D1H1S1	11	2	0	- 16	26	2	960	0.00
	R2D1H1S1T1	10	5	1	5	6	- 14	640	0.31
	R2D1H1S1T2	21	2 -	1	9	- 14	104	1920	5.63
	R2D1H1S2	10 -	1	16	- 3		- 10	2880	0.03
	R2D1H1S2T1	11	1	4	- 5		-114	1920	6.77
	R2D1H1S2T2	18	4 -	1	- 6	14	56	5760	0.54
	R2D1H2	5	4	17	- 13		-161	4320	6.00
	R2D1H2T1	5	11	5	- 13	- 37	- 5	2880	0.01
	R2D1H2T2	10	33	0	11	8	-533	8640	32.88
	R2D1H2S1	10	9	12	- 23	2	- 92	2880	2.94
	R2D1H2S1T1	9	5	10	20	• 5	- 74	1920	2.85
	R2D1H2S1T2	11	5	5	- 12	- 17	-320	5760	17.78
	R2D1H2S2	10	4 -	1	- 3	26	148	8640	2.54
	R2D1H2S2T1 R2D1H2S2T2	11	5	7	- 25	8	202	5760	7.08
		16	4 -	-	- 17	1	208	17280	2.50
22000	RZDZ	9	14 -	8	- 19	12	- 39	432	3.52

Table 39 .-- ANALYSIS OF VARIANCE USING ORTHOGONAL POLYNOMIALS (4).-- Continued.

			W. E. 192					
RDHST	Treatment	NO3-N values	I	11	ш	IN A	Divisor	Sum of squares
22001	R2D2T1	12	8	8	- 22	- 1 - 6	288	0.13
	R2D2T2	14	24 .	. 2	- 7	- 22 - 30	864	1.04
	R2D2S1	14	2	22	5	- 11 - 27	288	2.53
	R2D2S1T1	17 -	6	11	- 19	- 38 6	192	0.19
	R2D2S1T2	17	10 -	. 3	13	- 19 6	576	0.06
	R2D2S2	11	2	8	12	- 22 -135	864	21.09
	R2D2S2T1	22	4	14	- 15	27 30	576	1.56
	R2D2S2T2	24	9	7	- 1	- 37 - 18	1728	0.19
	R2D2H1	8	12 -	-	- 1	50 53	288	9.75
	R2D2H1T1	12	12 -	_	19	- 6 - 5	192	0.13
	R2D2H1T2	9	41 .		18	31 35	576	2.13
	R2D2H1S1	9	4 .	26	0	- 34 50	192	13.02
	R2D2H1S1T1	10	7	12	- 5	- 11 - 6	128	0.28
	R2D2H1S1T2	14	15 -	1	- 17	20 8	384	0.13
	R2D2H1S2	10	5 -	. 5	- 3	27 38	576	2.51
	R2D2H1S2T1	13	6	17	- 11	- 4 - 14	384	0.51
	R2D2H1S2T2	17	14 -		- 46	- 49 -52	1152	2.35
	R2D2H2	6 ~	10 -		29	23 - 39	864	1.76
	R2D2H2T1	7 -	7 -		- 12	- 26 - 87	576	13.14
	R2D2H2T2	11	2 -	-	5	34 33	1728	0.63
the body office and desire	R2D2H2S1	9	0 -		- 13	- 54 - 48	576	4.00
	R2D2H2S1T1	10 -	8 -	-	- 8	- 14 30	384	2.34
	R2D2H2S1T2	15 -	7 .	-	0	18 - 24	1152	0.50
	32D2H2S2	10 -	2 -		22	33 72	1728	3.00
	R2D2HUS2T1	13	1 .		4	- 22 174	1152	26.28
	R2D2H2S2T2	17	3 -	. 5	9	14 252	3456	18.38
23000		8	5	20	- 10	- 44 95	2160	4.18
The same of the sa	R2D3T1	13	3	3	34	31 40	1440	1.11
	R2D3T2	12 -	1 -	6	49	6 56	4320	0.73
	R2D3S1	12	12	1	8	14 - 15	1440	0.16
	R2D3S1T1	12	2 .	. ī	40	11 16	960	0.27
	R2D3S1T2	15	3	10	75	15 - 36	2880	0.45
	R2D3S2	21	3	10	53	- 36 101	4320	2.36
	R2D3S2T1	18	3	4	9	14 - 44	2880	0.67
	R2D3S2T2	31	6	4	7	- 1 - 64	8640	0.47
	R2D3H1	6	2	10	2	- 14 -179	1440	22.25
	R2D3H1T1	12 -	7	12	- 14	1 7	960	0.05
	R2D3H1T2	11	4	0	16	20 79	2880	2.17
	R2D3H1S1	10 -	2	3	- 1	- 1 - 46	960	2.20
	R2D3H1S1T1	11 -	1	11	11	36 - 98	640	15.01
	R2D3H1S1T2	4	5	5	12	- 9 - 52	1920	1.41
	R2D3H1S2	11 -	1	2	18	26 -110	2880	4.20
23121	R2D3H1S2T1	15 -	2	1	- 4	7 - 98	1920	5.00
	R2D3H1S2T2	23 -	2	0	13	- 41 112	5760	2.18
23200	R2D3H2	7 -	- 3 •	18	24	12 - 7	4320	0.01

Table 19 . -- ANALYSIS OF VARIANCE USING ORTHOGONAL POLYNOMIALS (4). -- Continued.

			otal or	ent of the American				
RDHST	Treatment	NO3-N	I	II	III	IV	V Divisor	Sum of
		values	_					squares
ARTER TOWNSHIPS	and the control of th	manda a community of the same	- disprints speni	teritor resignativo	SUPPLY SUPPLY			THE RESERVE OF THE PERSON OF T
23201	R2D3H2T1	7	0 -	25	12	- 20 2	5 2880	0.22
23202	R2D3H2T2	14	2 -	3	23	- 27 -27	1 8640	8.50
23210	R2D3H2S1	9	0 -	4	49	- 84 15	6 2880	8.45
23211	R2D3H2S1T1	10	7	3	38	83 8	2 1920	3.50
23212	R2D3H2S1T2	17	5 -	4	17	26 18	0 5760	5.63
23220	R2D3H2S2	10	3	25	- 15	- 11 - 6	4 8640	0.47
23221	R2D3H2S2T1	15 -	5	28	- 10	- 98 - 8	6 5760	1.28
23222	R2D3H2S2T2	24 -	1	5	- 28	117 -18	4 17280	1.96
30000		4	11	15	- 20	- 12 -21	1 2160	20.61
30001	R3T1	5 -	2 -	5	9	36 11	2 1440	8.71
30002	the second second	8	2	4	25	- 56 - 6		0.95
30010	The state of the s	7 -	3	40	10	87 7	7 1440	4.12
	R3S1T1	10 -	1 -	1	- 24	- 14 7		6.34
	R3S1T2	7	6 -	. 8	- 4	38 1		0.07
30020	The state of the s	6 -	1	28	1	24 32		25.06
	R3S2T1	12 -	4	15	52	116 20		14.17
	R3SZT2	28	0 -	1	26	156 9		1.11
30100		4	5	17	- 17	54 9		6.81
an an an an	R3H1T1	10	0	2	27	178 -13		17.88
	R3H1T2	6	5	1	- 10	132 - 1		0.08
	R3H1S1	7 -	1 .		10	11 - 5		2.82
	R3H1S1T1	9	2	9	- 3	47 - 5		4.56
	R3H1S1T2	12	0	1	- 1	48 - 2		0.41
	R3H1S2	10	0 -		0	69 - 8		2.45
	R3H1S2T1	9	2	1	- 7	14 - 9		5.00
	R3H1S2T2	12 -	1	3	- 12	16 -18		5.63
30200	THE RESERVE OF THE PARTY OF THE	8 -	4	6	4	8 - 2		0.14
the same and the same	R3H2T1	7	5 -	1	10	26 7		2.17
	R3H2T2	7	12	3	12	- 64 5		0.40
	R3H2S1	8 -	2	13	- 1	- 58 - 1	and the same of th	0.03
	R3H2S1T1	8 -	6	7	16	24 22		25.67
100 . 110 . 110 . 110 .	R3H2S1T2	9	2	8	18	- 6 4		0.34
	R3H2S2	10	0	1	17	57 7		9.63
	R3H2S2T1	8	1	5	58	- 3 3		0.20
	R3H2S2T2	14 -	1 -	5	33		4 17280	0.00
31000		6 -	7	20	38	- 27 40		15.19
and the second	R3D1T1		9 -	1	- 9	- 8 34		16.63
	R3D1T2	10	10	7	- 17	- 16 28		3.68
	R3D1S1 R3D1S1T1	6 -	2	14	- 1 5	- 24 19		5.28
	R3D1S1T2	17		20	2	- 76 18		7.36
	R3D1S2	12	3 -	3	- 1	- 54 - 2 - 66 -65		0.04
	R3D152T1	28	8	7	7	- 66 -65 4 -33		19.98
	Radiszii Radiszii	45	0	8	3	- 72 -74		7.65
44466	W-35-56-56	43	v	O	3	- 12 -14	9 43200	12.81

Table 19 .-- ANALYSIS OF VARIANCE USING ORTHOGONAL POLYNOMIALS (4).-- Continued.

ARTHUR LINES OF		A STATE OF THE STA	A COMMAND OF THE OWNER,				A STATE OF THE PARTY OF THE PAR
RDHST	Treatment	NO3-N I	II	III	IV V	Divisor	Sum of
		values					squares
					10 100	7000	
	R3D1H1	6 - 6	10	- 21	- 19 -179	7200	4.45
	R3D1H1T1	6 11	0	- 6	8 -281	4800	16.45
	R3D1H1T2	15 6	- 13		72 -341	14400	8.08
	R3D1H1S1	9 - 1	- 11	- 1	- 80 64	4806	0.85
31111		9 2	- 11	23	8 10	3200	0.03
	R3D1H1S1T2	14 9	9	37	- 16 112	9600	1.31
	R3D1H1S2	10 - 8	12	1	- 13 244	14400	4.13
4-4-1	R3D1H1S2T1	11 - 4	- 2	16	- 30 118	9600	1.45
	R3D1H1S2T2	15 - 3	0	31	8 64	28800	0.14
	R3D1H2	6 4	8	15	2 - 39	21600	0.07
	R3D1H2T1	5 11	29	- 13	- 51 73	14400	0.37
	R3D1H2T2	10 3	5	- 9	22 -315	43200	2.30
	R3D1H2S1	9 - 1	26	- 4	4 162	14400	1.82
31211		9 12	- 6	- 3	- 23 -274	9600	7.82
	R3D1H2S1T2	14 6	1	7	3 168	28800	0.98
	R3D1H2S2	10 5	15	7	- 70 - 30	43200	0.02
	R3D1H2S2T1	11 3	4	~ 3	- 16 106	28800	0.39
31222	R3D1H2S2T2	14 4	- 2	- 16	- 5 -432	86400	2.16
32000	R3D2	6 - 1	22	- 26	2 - 35	2160	0.57
32001	R3D2T1	6 - 3	24	- 3	- 85 190	1440	25.07
32002	\$3D2T2	20 - 9	3	- 13	34 70	4320	1.13
32010	R3D2S1	9 - 7	29	7	43 - 77	1440	4.12
32011	R3D2S1T1	15 3	5	- 51	- 36 62	960	4.00
32012	R3D2S1T2	17 1	7	- 40	- 79 - 38	2880	0.50
32020	R3D2S2	12 3	6	8	- 16 -413	4320	39.48
32021	R3D2S2T1	30 4	9	- 9	5 -302	2880	31.67
32022	R3D2S2T2	36 1	- 1	- 1	27 -218	8640	5.50
32100	R3D2H1	6 - 6	- 2	8	- 62 - 33	1440	0.76
32101	R3D2H1T1	11 3	11	13	26 -131	960	17.88
32102	R3D2H1T2	8 16	3	6	17 177	2880	10.88
32110	R3D2H1S1	10 - 7	20	- 14	116 60	960	3.75
	R3D2H1S1T1	11 - 8	5	- 14	9 - 42	640	2.76
	R3D2H1S1T2	4 4	1	0	92 84	1920	3.68
	R3D2H1S2	9 7	- 1	29	- 55 264	2880	24.20
	R3D2H1S2T1		9	- 14	30 190	1920	18.80
	R3D2H1S2T2	19 4	12	- 26	- 5 24	5760	0.10
	R3D2H2	8 2	- 5	29	- 93 -257	4320	15.29
	R3D2H2T1	8 - 6	5	34	- 54 - 53	2880	0.98
THE STATE OF THE STATE OF	R3D2H2T2	10 10	7	- 3	- 92 -233	8640	6.28
	R3D2H2S1	10 - 10	10	14	- 60 - 62	2880	1.33
	R3D2H2S1T1		- 5	- 21	- 8 26	1920	0.35
	R3D2H2S1T2		- 5	- 5	- 40 -140	5760	3.40
	R3D2H2S2		- 2	11	- 45 130	8640	1.96
	R3D2H2S2T1	14 1	12	7	- 52 274	5760	13.03
	R3D2H2S2T2		- 3	- 14	- 60 352	17280	7.17
3 60 60 60 60	271757179717	12 0	- 3	- T.4	- 00 332	11200	1 . 1. /

Table 90.--ANALYSIS OF VARIANCE USING ORTHOGONAL POLYNOMIALS (4).--Continued.

RDHST	Treatment	NO3-N value		II	III	IV	V	Divisor	Sum of squares
33000	R3D3	6	2	- 15	29	- 18	375	10800	13.02
	R3D3T1	9		- 7		61		7200	4.11
	R3D3T2	18	1	- 16			-396	21600	7.26
	R3D3S1	9	9	- 22		- 16		7200	0.42
	R3D3S1T1	22	5	4		- 31		4800	0.27
	R3D3S1T2	21	3	- 3			-188	14400	2.45
	R3D3S2	9		- 15		96		21600	0.57
33021	R3D3S2T1	29	5	- 19	39	- 34	- 44	14400	0.13
33022	R3D3S2T2	50	2	3	5	- 23	-228	43200	1.20
33100	R3D3H1	6	14	- 4	- 23	8	-253	7200	8.89
33101	R3D3H1T1	12 .	- 4	- 12	26		- 47	4800	0.46
33102	R3D3H1T2	10 -	- 12	- 4	- 13	44	173	14400	2.08
33110	R3D3H1S1	10 -	- 8	4	- 21	- 19	-152	4800	4.81
33111	R3D3H1S1T1	10 .		13		30	- 90	3200	2.53
3112	R3D3H1S1T2	17	4	- 1	29	147	124	9600	1.60
33120	R3D3H1S2	10	2	24	- 8	- 24	-112	14400	0.87
33121	\$3D3H1S2T1	13	2	- 8	15	33	-314	9600	10.27
33122	R3D3H1S2T2	25	1	7	- 29	103	~352	28800	4.30
	R3D3H2	6	6		- 17	96	147	21600	1.00
and the same of the same	R3D3H2T1	7 -	- 14	2		- 20	- 29	14400	0.06
	R3D3H2T2	11		- 2	26	167	-345	43200	2.76
	R3D3H2S1	10 -			16	- 30	154	14400	1.65
	R3D3H2S1T1	9	7	400 000	- 23	71	42	9600	0.18
	R3D3H2S1T2	16		- 1	47		- 44	28800	0.07
	R3D3H2S2	9	3		- 5	123	- 90	43200	0.19
	R3D3H2S2T1			- 13		40		28800	16.15
3222	R3D3H2S2T2	23	2	- 11	50	33	336	86400	1.31
Sub-to	tals	1209	L306	3208	1479	2340	4970		
		1567 1					-922		
		2081 2			1890		4818		
					2460		4498		
ub-to	tal sum	4857 5	885	7795	7577	11518	13364		
heck	total	5885 7	7795	7577	11518	13364			

APPENDIX II

Analysis of Variance taken from Snedecor (26)

Coded Data

herbicide						1						,					2)											0	3					
herbicde rate		0		1			2			3		(0			1			2			3		($\overline{\bigcirc}$			1			2			3	
soil type	1	2 3	3 1	2	3	1	2	3	1	2	3	1	2	3	1,	2	3	1	2	3	1	2	3	1	2	3	1	2	3	1	2	3	1	2	3
sampling date	68 10 10	19 11 19 19 19 19 19 19 19 19 19 19 19 1	14 11	0000	1 19	10 9 14 12	14 21 17	20 19 44 31	10 10	17 17 21	28 45 36 50	68 89	8 /2 /4	/! 8 9 35	777011	9 12 6 15	14 17 18 25	67911	11 21 14	17 18 17 23	6 15 8 10	12 14 17	15 19 25	8 9 9	13 14 15	16 16 19 24	9 09 9	10 11 14	13 16 19	7 10 11 14	13 11 15	1677	7 10 10	9 14 14 16	14 19 23
																4	2.	5																	
sampling date 2	9 69 10	13 /	10 12 12 15	0 6	1 10 12 11 0 15	10 6 12 13	10 17	13 22 18	5 7 69	10 11 15 22	12 29 30 39	54 9 9	12 10 14 10	12 12 15	77910	9 11 10	9 14 13 16	87/2	9 10 10 11	9 11 13 15	10 611 18	9 11 10	9 11 12 15	7697	109 13	10157	7 8 9 8	8 4 11 9	10 13 15 17	9 5 7 7	109 100	12 11 13 15	7 5 8 7	9 11 9	8 11 14 15
																	3	5																	
sampling date 2	2987	7 9 9 8	9 9	3 4 7 1 8 1 5	8 10 12 12	9 9	/3 /0 /4 /2	14 10 11 21	4 6 6 6	7 6 9 9	6 12 129	12 9 6	109 10	10 10	4667	8 10 9	10 11 10 10	9 6	9 11 9 10	10 10 11	4666	7 9 10 10	10 10 10	4687	9 10 10	9 10 9	4676	2990	7 9 10 9	3 6 7	10 9 9	9 10 10 10	8 6 9 6	9 10	10 10 10 9

V		77
\mathcal{N}	匚	I

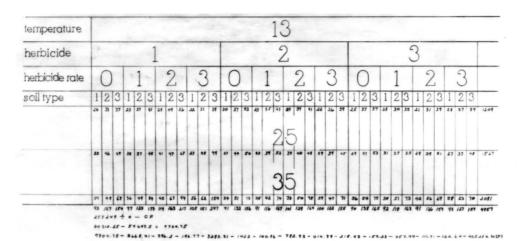
tempercture	herbicide	her	bicio	de 1	ate			soil type	sampling date
1 13°C	1 CaCN ₂	herbicide	0	1	2	3	1	Gilcrest sand	1 5 days
2 25°C 3 35°C	2 Chloro-IPC	1 0 1	check		200	400	2	Rocky Ford loam	2 10 days
3 35 C	3 444E	0	check check		3+3	6 20	3	Las Animas clay	3 20days 4 30days
		1 All he expresse herbicide	ed as	pour	nds o				

25 35 1 1 2 3 3 1 1 2 3 3 1 1 2 3 3 1 1 2 3 3 1 1 2 3 3 1 1 2 3 3 1 1 2 3 3 1 1 2 3 3 1 1 3 1 1 1 1
3

Main effects of temperature, soil type, and herbicides.

57				her	bici	ide :	rate								SC	am	olir	ng (date	9				;
	0	-		1			2			3			1			2	1		3			4		
29 37 7 9 9 9 9 9 9 9 9 12 8 6 8 10 9 10 4 9 10 10 9 6 8 7 8 9 10 10 9 10 9 9	964 03049 0225549 92020112225769 7 093511057	69 10 10 12 14 12 12 14 14 14 14 18 18 18 18 18 18 18 18 18 18 18 18 18	378561128801211466788090110046762991079109	10611119991010211577910991109431678889119103157	60000071121492977011926547959090011463695	4799304240112969691191010101356731999101010	10623817244328872229101191135957710910121135	109411115099431679111214478773701143115721674	4666769962294666790000908686890000009	5769 10115228809 1061129 9 110 9 11257 587 89 119 8 1145	8 10 0 8 7 7 7 7 7 1 2 5 6 6 6 6 5 8 10 2 1 4 7 7 2 5 9 6 7 7 7 10 11 9 4 4 16 4 4 19 23	27 03 68 43 14 7 6 4 8 4 4 8 0 9 9 10 4 7 10 4 8 9 4 27 5 3 9 8 8 10	931009001450252117999999997011790902799	62260404097969117946117622836903732794	9997110700662220968116110690690699500690	610,269,260,37 1/24 10,27 9 4 7 1011 69 11 4 9 10 893 591 591	81440799190755718727772854592601160116044	89 98 12 12 9 14 11 6 9 12 8 9 10 6 10 8 9 10 6 10 9 8 10 9 7 9 10 6 9 10 9 10 9 10 9 10 9 10 9 10 9	84111911217265397229113203111293581157038114	102901124742736849068944784994994911571044	789581182269960079 0600160070960979 0609	19 1 5 1 1 10 5 3 2 2 9 9 2 2 9 9 15 10 10 10 12 11 5 2 10 3 7 5 17 8 9 7 7 7 15 7 9 5	10 103 10 12	
495 COT	379 Tect	473	288 facto	370 Or (492	333 ÷ 43	400 J =	531	193	419	595	246	334	397	3/3	345			436	528	320	452	650	
		700	5480		T13 /)	÷ 45	∀ =	3460	<i>1.</i> 5				05 945 610.6				100							
5480	4. 27 -	54607.	5 = 1	96.77											100									

Main effects for herbicide rates and sampling dates.



Š

temperature						1	6.5	3										2	5											3	5)					
herbicide rate	-	C)		1			2	2		3	,		0			1			2			3	,	(0			1			2	,		3		
soil type	1	2	3	1	2	3	1	2	3	1	2	3	1	2	3	1	2	3	1	2	3	1	2	3	1	2	3	1	2	3	1	2	3	1	2	3	
sampling date		107	23 28 28 28	11 19 21 18	16 38 31 27	35 30 32 30	16 18 23 21 78	25 31 32 31	33 30 31 42	18	24 29 29 104	34 31 28	21 16 26 26	35 29 31 31 31	32 34 39 47	24 21 29 29	26 27 31 29	39 39 48 155	27 18 31 32	27 29 37 33	35 35 48 48	29	41	56	20 25 27 32 104	40	48 51 78	31	30 31 51	59	34	53 46 36	49 53 58 78 238	38	18 45 35 54	98	977 1164 1294 1422 485

1923 11+3-CF

64 103- 67 - 54 607.5 = 9496.17

9496.17-266.41-146.77-3283.31-1003.10-140.46-78243-323.63-157.23-56.30-145.06-120.64-203.48-117.06-129.44=117.25

117.25

herbicide						-	1											2	2												3						
herbicide rate	(0			1			2			3		1	0		1	1			2			3		(0			1			2			3		
soiltype	1	2	3	1	2	3	1	2	3	1	2.	3	1	2	3	1	2	3	1	2	3	1	2	3	1	2	3	1	2	3	1	2	3	1	2	3	
sampling date 2	17 23 26 27	32 33 35	32 35 40	21 21	25	41 45	35	35	48 42 57 70	17 23 32 33	24 34 41 52	#L #5 78	15 24 25 27	28 32 35 37	39	18 20 25	26 29 27 34	33 42 41 51	23 20 29 29	29 42 33	36 39 40 49	20 27 25 29	29 32 45 37	36 40 48	19 21 26 24	31 30 37 40	36 36 43 80	20 24 24 24	20 29 34 35	30 38 49 51	19 10 24 28	26 30 34 36	33 37 40 49	21 26	35 35 35	35 43 47	977 164 294 412

184449 + 3 - CF

6483.0 - 5+607.5 = 6875.5

6875.5-346.2-146.77-3283.31-1003.10-414.39-218.93-65.62-157.23-56.30-195.06-405.25-180.90-42.70-117.06=142.69

142.68

temperature					15	3							2		5						,	-	35	5				
herbicide		1			2	1		3			1			2			3			1			2			3		
soil type	1	2	3	1	2	3	1	2	3	1	2	3	1	2	3	1	2	3	1	2	3	1	2	3	1	2	3	
sampling date	31	33 34 44 37	35 41 49 50	30 49	34 37 39 39	14 40 37 4/	19 23 29 26	37 38 39	35 89 39 37	34 25 37 43	40 55 53	46 65 75 77	30	39 39 44	31 49 50 59	30 24 31 29	36 45 43	4 45 53	30 37 54 50	43 59 57 67	74 97 /6/ 133	25 37 35	40 54 37 63	54 68 73 98	31 39 39 45	45 48 57 64	55 (2 74 96	977

256713 + 4 - CF

64179.25 - 54607.5 = 9570.75

9570.75 - 346.2 - 2668.41 - 3283.31 - 1003.1 - 141.20 - 782.43 - 325.63 - 218.43 - 68.62 - 148.06 - 111.71 - 28.76

temperature.		1	3			2	5				35	5	
herbicide rate	0	1	2	3	0	1	2	3	0	1	2	3	
sampling date	56 83 80 76	52 77 84 75	74 79 86 94	64 74 80 75	79 104 107	79 17 98	84 82 116 113	78 97 118 136	92 111 124 146	92 109 119 163	110 132 138 151	/03 54 47 91	977 1164 1294 1422

temperature		13	3			2	5			3	35		
herbicide rate	0	1	2	3	0	1	2	3	0	1	2	3	
soiltype	8/2 /07	69 102 117	78 119 136	72 /04 //7	89 137 152	102 113 155	101 126 166	93 134 192	104 153 216	109 142 231	/20 /73 238	/33 /62 300	1572
	195	288	393	293	378	370	400	419	473	482	531	595	485

532/35 + 9 - CF

59126.11 - 54607.5 = 4518.61

4518.61-2668.41-196.77-1003.10-323.63-56.30-140.96= 124.44

129.4

743493 ÷ 12 - CF

61957.75 - 54607.5 = 7350.25

7350.25-2668.41-196.77-3283.31-782.93-140.96-157.23= 120.64

120.64

herbicide		1			2			3		
soiltype	1	2	3	1	2	3	1	2	3	
herbicide rate	93 97 114 105	127 123 163 151	154 173 217 297	91 91 100	132 116 129 /42	156 167 164 155	90 92 91 93	138 118 126 127	163 163 159 157	1/46 1/40 1/264 1/307
*	409	564	841	383	499	642	344	5 09	644	40

716155 + 12 - CF

59 679. 58 - 54 607. 50 = 5072.08

5072.08 - 396.20 - 3293.31 - 196.77 - 218.93 - 414.39 - 157.23 = 405.25

405.25

herbicide			1			2	2		3				
herbicide rate	0	1	2	3	0	1	2	3	0	1	2	3	
sampling date	101	76 91 106 120	107 105 140 143	87 142 151 173	69 95 101 114	77 91 93 113	94 01 02 03	79 95 90 113	86 87 106 114	70 91 102 110	78 87 98 //3	79 88 104 106	977 /64 1194 422

512 287 ÷ 9 - C F 56920.78 - 546.07.5 = 2313.28

2313.28-396.20-196.77-1003.10-414.39-65.62-56.30=180.90

180.90

soil type]				4	2		3				
herbicide rate		1	2	3	0	1	2	3	0	1	2	3	-
sampling date	51 68 77 78	78	66 62 88 90	83 85	91 95 107 104	93	90 /13 /15 /00	77 98 101 124	94 110 124 147	95 121 130 157	117 18 137 168	109	977

536547 + 9 - CF

59616.33 - 54607.5 = 5008.83

500 9.93 - 32 83.31 - 196.77 - 1003.10 - 157.23 - 195.06 - 56.30 = 117.06

117.06

temperature		13			25			35		
soil type	1	2	3	1	2	3	1	2	3	
sampling date	53 84 81 77	111 120	/07 120 122 /28	94 73 110 115	/15 /14 /44 /37	125 158 182 200	86 113 128 139	128 166 152 184	183 227 248 327	977 64 294 422

7568.5 \div /2 - C F 63067.92 - 54607.5 = 8460.42 84 60.42 - /003/0-328331-2668.41 - 782.95-323.63-185.06 = 203.98 203.98

temperature		13			25)		35 1 2 3		
herbicide	1	2	3	1	2	3	1	2	3	
sampling date	1 84 2 106 3 119	87 108 105	75 99 106	130	107 110 135 143	107	147 193 212 250	119	131 149 170 205	977 1164 1294 1482
	422	405	382	590	495	482	802	624	655	485

532/35 + 9 - CF 59/26.11 - 54607.5 = 45/8.61 45/861 - 2668.41 - 196.77 - 1003.1 - 323.63 - 56.3 - 140.96 = 129.44 129.44

temperature herbicide	1	13	3	1	2	3	1	2	3	
soil type	150	104	97 135 150	139	138	115	171 226 405	141	154	1158

995379 + 16 - CF

62211.19 - 54607.5 = 7603.69

7603.69-2668.41-396.2-3213.31-782.93-142.2-218.93=11.71

111.7

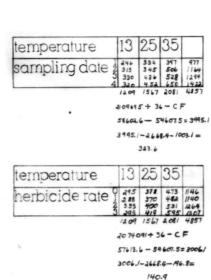
temperature		13		1	25)	,	35		
herbicide	1	2	3	1	2	3	1	2	3	
herbicide rate	101	100	99 88 91	128 123 155 184	121 124 127 123	129 123 118 112	150	158 151 158 157	165 162 167 /61	1146 1140 1264 1307
1	422	405	385	590	445	482	802	624	655	4857

 $707093 \div 12 - CF$ 58924.42 - 54607.5 = 4316.92 4316.92 - 140.96 - 140.96 - 140.96 = 35299 357.99

herbicide ::		1			2			3		
soil type	1	2	3	1	2	3	1	2	3	
samplingdate	77 97 122 119	116 135 156 157	158 203 220 260	76 91 104 113	111 135 140 133	126 156 163 198	80 86 100 100	121	/3/ /46 70	977
	409	564	841	383	499	641	366	509	644	485

7/7749 + 12 - CF 59812.43 - 54607.5 = 5204.92 5204.92 - 396.20 - 3283.31 - 1003.10 - 2/9.93 - 65.62 - 195.06 = 42.7

42.7



0 374 379 393 //46 1 393 374 313 //40 2 494 394 376 /a64 3 553 377 377 607

2002135 + 36 - CF

55614.9-54607.5=1007.4

1007.4-396.2-196.8=

414.4

herbicide

herbicide rate

			/95.	1	
herbicide		1	2	3	
soil type	1 2 3	409 564 841	313 498 642	366 509 644	15
					_

480	8285 + 48 - CF
51	505.9 - 54607.5 = 3898.4
384	8.4-396.2-3283.3=
	218.9

herbicide rate	0	1	2	3	
sampling date	236 273 308 329	243 273 301 343	273 293 340 358	245 325 345 392	977
	1146	1140	1264	1307	4857

508319 + 47 - CF 55863.7-54607.5=1256.2 12562-196.8-1003.1=

54.3

2-way tables

				1.5					
mperature	13	25	35		temperature	13	25	35	
il type	1 300 2 43 3 47	392 5 10 7 665	466 630 985	1158 1572 2127	herbicide	1 422 2 405 3 314	590 495 482	802 624 655	1.
	140	9 156	2081	4857		12.09	1567	2081	4

2944423 + 48 - CF 61342.1-54607.5=6734.6 67346-2669.4-3283.3= 784.9

soil type	1	2	3	
sampling date	233 248 334 331	329 391 416 436		977 1164 1294 1423
	1158	1572	2/27	405

2/27203+36-CF
5908 8.9-54607.5= 448/.5
4481.5-3283.3-1003/=

	1814	1524	1519	4857
soiltype	409 564 841	313 498 642	366 509 644	1158 1572 2127
herbicide	1	2	3	

The second second second	
4808285 + 48 - CF	
51505.9 - 54607.5 = 3898.4	
3848.4 - 396.2 - 3283.3 =	

ing date	236	223 273 301	273 293 340	245 325 345	977
ide rate	0	1	2	3	

	2775087÷48 - CF			
	1209	1567	2081	485
herbicide	1 422 2 405 3 312	590 495 481	802 624 655	1524
lemperdiare			00	

57814.3-54 607.5=3206.8
3206.8-2668.4-396.2=
142.2

herbicide	1	2	3	
sampling date	351 429 429 498 536	3/3 382 384 443	3 /3 3 5 3 4 / 0 4 4 3	977 1164 1294 1422
	1814	1534	1510	405

2018607+ 36 - CF 56072.4 - 54607.5= 1464.9 1464.9-396.2-10031= 65.6

soil type	1	2	3	
herbicide rate	274 280 306 298	397 357 418 400	475 503 540 609	1146 1140 1264 1307
	1156	1572	2127	4857

2096813:36-CF 58244.8-54607.5=3637.3 3637.3-3283.3-196.8= 157.2

Table 12. -- ANALYSIS OF VARIANCE.

	Andrew Company			
	D/F	SS	_M 2	erper
Temp	2	2,668.41	1,334.21	251.74**
Dates	3	1,003.10	334.37	63.08**
Soils	2	3,283.31	1,641.65	309.75**
Herb	2	396.20	198.1	37.38**
Rates	3	196.77	65.59	12.37**
		7,547.79		
Temp x dates	6	323.63	53.94	10.18**
Temp x soils	4	782.93	195.73	36.93**
Temp x herbs	4	142.20	35.55	6.71**
Temp x rates	6	140.96	23.49	4.43**
Dates x soils	6	195.06	32.51	6.13**
Dates x herbs	6	65.62	10.94	2.06*
Dates x rates	9	56.30	6.25	1.18
Soils x herbs	4	218.93	54.73	10.34**
Soils x rates	6	157.23	26.20	4.94**
Herbs x rates	6	2,497.25	69.06	13.03**
Temp x dates x soils	12	203.98	16.99	3.21*
Temp x dates x herbs	12	28.76		
Temp x dates x rates	18	129.44	7.19	1.35
Temp x soils x herbs	8	111.71	13.96	2.63*
Temp x soils x rates	12	120.64	10.05	1.98*
Temp x herbs x rates	12	357.99	29.83	5.63**
Dates x soils x herbs	12	42.70	3.55	
Dates x soils x rates	18	117.06	6.50	1.22
Dates x herbs x rates	18	180.90	10.05	1.89*
Soil x herb x rates	12	405.25 1,698.43	33.77	6.37**
Temp x herb x rates x soils	24	307.83		
Temp x herb x soils x dates	24	104.21		
Temp x herb x rates x dates	36	184.83	8	
Temp x rate x soils x dates	36	117.25		
Herb x rates x soils x dates	36	142.68		
1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1		856.80		
Temp x dates x soils x herb				
x rates	72	351.23		
ERROR	228	1,208.03		
Total	431	12,951.50		

BIBLIOGRAPHY

BIBLIOGRAPHY

- Aldrich, R. J. Herbicides; residues in soil. Journal of Agricultural and Food Chemistry, 1:257-260, April 1953.
- Audus, L. J. Biological detoxication of 2,4-dichlorophenozyacetic acid in soils: Isolation of an effective organism. Nature, 166:356, August 26, 1950.
- 3. Audus, L. J. Plant and Soil, 3:170-192, 1951. Summarized in Aldrich (1:259).
- 4. Bainbridge, J. R., Grant, A. M., and Radok, U.
 Tabular analysis of factorial experiments and
 the use of punch cards. American Statistical
 Association. Journal, 51:149-158, March 1956.
- 5. Carlyle, R. E., and Thorpe, J. D. Some effects of ammonium and sodium 2,4-dichlorophenoxyacetates on legumes and the Rhizobium bacteria.

 American Society of Agronomy Journal, 39:929-936, October, 1947.
- Cowie, G. A. Decomposition of cyanamide and dicyanodiamide in the soil. Journal of Agricultural Science, 9:113-136, April 1919.
- 7. Crowther, E. M., and Richardson, H. L. The decomposition of calcium cyanamide in the soil
 and its effects on germination, nitrification
 and soil reaction. Agri-Chem Review, 1:315, December 1952.
- 8. De Rose, H. R. Persistence of some plant growthregulators when applied to the soil in herbicidal treatments. Botannical Gazette, 107:583-589, June 1946.
- 9. De Rose, H. R., and Newman, A. S. The comparison of the persistence of certain plant growth regulators when applied to soil. Soil Science Society of America. Proceedings, 12:222-226, 1948.

BIBLIOGRAPHY. -- Continued

- 10. Fults, J. L., and Payne, M. G. Some effects of 2, 4-D, DDT, and Colorado 9 on the bacteria

 Rhizobium leguminosarium Frank in the root nodules of the common bean. American Journal of Botany, 34:245-248, May 1947.
- 11. Jones, H. E. The influence of 2,4-dichlorophenoxyacetic acid on nitrate formation in a prarie soil. American Society of Agronomy. Journal, 40:522-526, 1948.
- 12. Koike, H., and Gainey, P. L. Effects of 2,4-D and CADE, singly and in combination, upon nitrate and bacterial content of soils. Soil Science, 74:165-172, August 1952.
- 13. Kries, O. H. Persistence of 2,4-D in soil. Botanical Gazette, 108:510-525, 1947.
- 14. Lewis, R. W., and Hammer, C. L. The effect of 2,4-D on some microorganisms. Michigan Agricultural Experiment Station Quarterly Bulletir, 29:112-114, November 1946.
- 15. Loustalot, A. J., and Ferrer, R. The effect of some environmental factors on the persistence of sodium pentachlorophenate in the soil.

 American Society of Horticulture Science.

 Proceedings, 56:294-298, December 1950.
- 16. Mukerji, B. K. Microbiological aspects of nitrification in soils under varied environmental conditions. Agri-Chem Review, 1:3-8, April 1953.
- 17. Newman, A. S. The effect of certain plant growth regulators on soil microorganisms and microbial processes. Soil Science Society America. Proceedings, 12:217-222, 1947.
- 18. Newman, A. S., and Thomas, J. R. Decomposition of 2,4-dichlorophenoxyacetic acid in soil and liquid media. Soil Science of America. Proceedings, 14:160-164, 1949.
- 19. Newman, A. S., De Rose, H. R., and De Rigo, H. T.
 Persistence of isopropyl N-phenyl carbamate in
 soils. Soil Science, 66:393-397, November 1948.

BIBLIOGRAPHY, -- Continued

- 20. Newman, A. S., Thomas, J. R., and Walker, R. L.
 Disappearance of 2,4-dichlorophenoxyacetic acid
 and 2,4,5-trichlorophenoxyacetic acid from soil.
 Soil Science Society of America. Proceedings,
 16:21-24. January 1952.
- 21. Norman, A. G., and Newman, A. S. Proc. Northeastern weed cont. conf. 1950, 7-14. Summarized in Aldrich (1:258).
- 22. Priola, M. A. Chemical weed control in onions.

 Master's thesis, 1954, Colorado A & M College,

 97p., typewritten.
- 23. Ries, S. K. Chemical weed control in spinach.

 Master's thesis, 1951, Cornell University,
 typewritten.
- 24. Schmidt, E. L. Soil microorganisms and plant growth substances. Soil Science, 71:129-140, February 1951.
- 25. Smith, N. R., and Dawson, V. T., and Wenzel, M. E.
 The effect of certain herbicides on soil microorganisms. Soil Science Society of America.
 Proceedings, 10:197-201, 1945.
- 26. Snedecor, G. W. Statistical Methods. Ames, Iowa, The Iowa State College Press, 1946, 485p.
- 27. Stevens, L. F. Effects of chloro-IPC on various crops and residual action of chemical when applied to various soils. Master's thesis, 1951, Michigan State College, typewritten.
- 28. Stevenson, E. C., and Mitchell, J. W. Bacteriostatic and bactericidal properties of 2,4-dichloro-phenoxyacetic acid. Science, 101:642-644, June 22, 1945.
- 29. Temme, Johannes. About the decomposition of Cacyanamide in the soil. n.p., 1946. 98p, typewritten.

 Dr. Agr. Thesis, 1946, The Agricultural College of Waganingen, translated by H. Hartjens. Copy in library of American Cyanamid Company. Agricultural Chemicals Division, 30 Rockefeller Plaza, New York 20, New York.

BIBLIOGRAPHY. -- Continued

- 30. Young, H. C., and Carroll, J. C. The decomposition of pentachlorophenol when applied as a residual pre-emergence herbicide. American Society of Agronomy. Journal, 43:504-507, October 1951.
- 31. ZoBell, C. E. Action of microorganisms on hydrocarbons. Bacteriological Reviews, 10:1-49, March-June, 1946.

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