How important is organic aerosol hygroscopicity to aerosol indirect forcing?

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How important is organic aerosol hygroscopicity to aerosol indirect forcing?

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Abstract
Organics are among the most abundant aerosol components in the atmosphere. However, there are still large uncertainties with emissions of primary organic aerosol (POA) and volatile organic compounds (VOCs) (precursor gases of secondary organic aerosol, SOA), formation of SOA, and chemical and physical properties (e.g., hygroscopicity) of POA and SOA. All these may have significant impacts on aerosol direct and indirect forcing estimated from global models. In this study a modal aerosol module (MAM) in the NCAR community atmospheric model (CAM) is used to examine sensitivities of aerosol indirect forcing to hygroscopicity (represented by a single parameter ‘κ’) of POA and SOA. Our model simulation indicates that in the present-day (PD) condition changing the ‘κ’ value of POA from 0 to 0.1 increases the number concentration of cloud condensational nuclei (CCN) at supersaturation \(S = 0.1\%\) by 40–80\% over the POA source regions, while changing the ‘κ’ value of SOA by \(±50\\%\) (from 0.14 to 0.07 and 0.21) changes the CCN concentration within 40\%. There are disproportionally larger changes in CCN concentration in the pre-industrial (PI) condition. Due to the stronger impact of organics hygroscopicity on CCN and cloud droplet number concentration at PI condition, global annual mean anthropogenic aerosol indirect forcing (AIF) between PD and PI conditions reduces with the increase of the hygroscopicity of organics. Global annual mean AIF varies by 0.4 W m\(^{-2}\) in the sensitivity runs with the control run of -1.3 W m\(^{-2}\), highlighting the need for improved understanding of organics hygroscopicity and its representation in global models.

Keywords: aerosol, organic, cloud, climate

1. Introduction

Atmospheric aerosols affect the global energy budget by scattering and absorbing sunlight (direct effects) and by changing the microphysics, lifetime, and coverage of clouds (indirect effects). An increase in aerosol concentration would lead to smaller cloud droplet size and higher cloud albedo, i.e., brighter clouds (Twomey 1977). This effect, which is known as the first indirect aerosol effect, tends to cool the global climate. The smaller cloud droplet size resulting from increased aerosol concentration also inhibits precipitation, leading to an increase in cloud lifetime and coverage (second indirect aerosol effect, Albrecht 1989). Although it is widely accepted that the indirect effects can strongly influence the global climate, and potentially mask the warming effect due to anthropogenic CO\(_2\), the magnitudes of indirect aerosol effects are poorly understood. The Intergovernmental Panel on Climate Change (IPCC 2007) considers the indirect effects of aerosol the most uncertain components in forcing of climate change over the industrial period.

To understand the aerosol indirect effects requires knowledge of the cloud condensation nuclei (CCN) spectrum of aerosol particles, which is the number concentration of
particles that can form cloud droplets as a function of water vapor supersaturation. The hygroscopicity of particles containing only soluble inorganic species typically found in atmospheric aerosols is well understood, and the CCN spectrum of the inorganic particles can be effectively predicted using Köhler theory (Köhler 1936). However, the composition of atmospheric particles is often very complex. Compared to the inorganic component, usually limited to a few species, the organic component of ambient aerosols may consist of hundreds or even thousands of species, which often contribute substantially to, or sometimes even dominate fine aerosol mass (Kanakidou et al 2005). A number of studies have shown that organic aerosols play an important role in both the direct and indirect aerosol forcing (e.g., Lohmann et al 2000, Chung and Seinfeld 2002, Kanakidou et al 2005, Goto et al 2008), which are highly uncertain (IPCC 2007). Direct radiative forcing of anthropogenic organic aerosol was estimated to be $-0.06$ to $-0.23$ W m$^{-2}$ as compared to $-0.16$ to $-0.58$ W m$^{-2}$ for anthropogenic sulfate from nine global models with the same prescribed emissions (Schulz et al 2006). At present there are large uncertainties in the emission inventories of primary organics aerosols and secondary aerosol formation in global models (Kanakidou et al 2005, Farina et al 2010). Furthermore, unlike typical inorganic aerosol species whose hygroscopicities are well understood, currently there are still large uncertainties in our understanding of the hygroscopicity of the organic species. Previous studies show atmospheric organics ranging from non-hygroscopic, such as primary organic aerosol emitted from fossil fuel combustion, to highly hygroscopic biomass burning organic aerosols (Carrico et al 2008, Petters et al 2009). Given the computational constraints, the large number of organic species often can only be represented by no more than a few classes in global models. This presents further challenges in simulating aerosol indirect forcing (AIF). In this study, a version of the NCAR Community Atmospheric Model (CAM) is used to examine the uncertainties in global simulated CCN concentration, cloud droplet number concentration (CDNC), and AIF arising from the uncertainty in our understanding of organic aerosol hygroscopicity. In the CAM, the activation of aerosol particles is represented using a parameterization (Abdul-Razzak and Ghan 2000) based on Köhler theory, which directly takes into consideration the effects of particle size, composition, and hygroscopicity of participating species (including organics) on particle activation. As far as we know, this is the first study examining the uncertainty in global averaged AIF due to the uncertainties in organic hygroscopicity. The implication on future study of the hygroscopicity of organic aerosol is also discussed.

2. Model

We use the NCAR Community Atmospheric Model (CAM) for all the simulations. This model version starts with CAM version 3 (Collins et al 2006) with several important changes added. The deep convection scheme closure has been modified following Neale et al (2008). The shallow convection scheme is based on Park and Bretherton (2009), and the moisture boundary layer scheme is based on Bretherton and Park (2009). The radiation code has been updated to the rapid radiative transfer model for GCMs (RRTMG) as described by Iacono et al (2008). The liquid cloud macrophysics closure is described by Park et al (2010). This CAM version is designed to simulate fully interactive aerosol direct and indirect effects. For that, a two-moment stratiform cloud microphysics is included which predicts the mass and number mixing ratio of cloud water and cloud ice, and diagnoses the mass and number mixing ratio of precipitation (rain and snow) (Morrison and Gettelman 2008). The cloud ice microphysics was further modified to allow ice supersaturation and aerosol effect on ice clouds (Gettelman et al 2010).

The new modal aerosol module (MAM) is based on Easter et al (2004) but includes many updates for aerosol treatment and has a simplified three-mode version added for long-term climate simulations (Liu et al 2010). The benchmark version of MAM with seven aerosol modes (MAM-7) is used here. It includes Aitken, accumulation, primary carbon, fine dust and sea salt and coarse dust and sea salt modes. The predicted internally mixed aerosol species for each log-normal mode in MAM-7 are listed in table 1. In MAM, the interstitial aerosol particles that are suspended in air (either clear or cloudy air) and the aerosol particles in stratiform cloud droplets (referred to as cloud-borne aerosol particles) are both explicitly predicted, as in Easter et al (2004). The cloud-borne aerosol particles are not transported, which saves computer time but has little impact on their predicted values (Ghan and Easter 2006). Below we will provide more description on the treatment of organic aerosol and aerosol activation most relevant to this study. Further detailed description on aerosol processes and properties treated in MAM can be found in the CAM scientific document (Neale et al 2010).

In MAM, the organic species are grouped into two classes. The first class, named as primary organic aerosol (POA), includes primary organics emitted due to both fossil fuel and bio-fuel consumption (industry, energy, transportation, agriculture waste burning, waste treatment, domestic emission and shipping) and biomass burning (forest fire and grass fire) (Lamarque et al 2010). The second class, referred to as secondary organic aerosol (SOA), includes secondary organics formed from anthropogenic and biogenic precursors. Carbonaceous aerosols (POA and black carbon (BC)) are emitted in the primary carbon mode (see table 1) and age to the accumulation mode through condensation of sulfuric acid gas (H$_2$SO$_4$), ammonia (NH$_3$), and secondary organic aerosol gas (SOA (gas)) and coagulation with Aitken and accumulation mode. The simplest treatment of SOA, which is used in many global models, is to assume fixed mass yields for anthropogenic and biogenic volatile organic compounds (VOCs), then directly emit this mass as primary aerosol particles. MAM adds one additional step of complexity by simulating a single lumped gas-phase SOA (gas) species. Fixed mass yields with values of 5%, 5%, 15%, 4% and 25% are assumed for big alkanes, big alkenes, toluene, isoprene and monoterpenes, respectively of the five VOC categories in the MOZART-4 gas-phase chemical mechanism (Emmons et al 2010). References for these yields can be found in Neale et al
The total yielded mass is emitted as the SOA (gas) species. MAM then calculates condensation/evaporation of the SOA (gas) to/from aerosol modes (Aitken and accumulation mode). Organic aerosol can affect stratiform cloud microphysics through participating in activation. The hygroscopicity of each component (including POA and SOA) is prescribed in MAM, and the volume mean hygroscopicity for each aerosol mode containing different components is calculated (Ghan and Zaveri 2007). Aerosol activation is parameterized in terms of sub-grid cloud scale vertical velocity, size distribution and volume mean hygroscopicity of all of the aerosol modes (Abdul-Razzak and Ghan 2000). The updraft velocity is approximated by the square root of the turbulence kinetic energy, and with a minimum value of 0.2 m s\(^{-1}\). A peak supersaturation is diagnosed in the Abdul-Razzak and Ghan (2000) parameterization which determines the critical size of aerosols in each mode activated to become cloud droplets. We note that currently the effects of organic aerosol on convective and ice cloud microphysics are not included in the CAM.

The hygroscopicities of POA and SOA are parameterized using the B value (Hudson and Da 1996), which is practically equivalent to the single parameter \(\kappa\) proposed by Petters and Kreidenweis (2007). In this study, the single parameter \(\kappa\) is used, as many recent measurements of organic hygroscopicity are reported in \(\kappa\). Table 2 lists the \(\kappa\) values of secondary organics aerosols and biomass burning organic aerosol reported from previous studies, including field measurements and environmental chamber studies. While fossil fuel primary organics are generally considered as non-hygroscopic (i.e. \(\kappa = 0\)), previous studies show that biomass burning organic aerosol and secondary organic aerosols are generally hygroscopic with \(\kappa\) ranged from 0.06 to 0.30. In addition, previous field studies suggest that the hygroscopicity of SOA increases from \(-0\) to 0.2 during its ageing in the atmosphere (Jimenez et al 2009).

To test the effects of hygroscopicity of organic aerosol on CCN and indirect forcing, we run the CAM by varying \(\kappa\) values from previous studies, including field measurements and environmental chamber studies. While fossil fuel primary organics are generally considered as non-hygroscopic (i.e. \(\kappa = 0\)), previous studies show that biomass burning organic aerosol and secondary organic aerosols are generally hygroscopic with \(\kappa\) ranged from 0.06 to 0.30. In addition, previous field studies suggest that the hygroscopicity of SOA increases from \(-0\) to 0.2 during its ageing in the atmosphere (Jimenez et al 2009).

### Table 1. Predicted species for interstitial and cloud-borne component of each aerosol mode in MAM-7. (Note: mmr—mass mixing ratio.)

<table>
<thead>
<tr>
<th>Organics class</th>
<th>Aitken mode</th>
<th>Accumulation mode</th>
<th>Fine soil dust</th>
<th>Fine sea salt</th>
</tr>
</thead>
<tbody>
<tr>
<td>Sea salt mmr</td>
<td>Number mixing ratio</td>
<td>Sulfate mmr</td>
<td>Sea salt mmr</td>
<td>Sea salt mmr</td>
</tr>
<tr>
<td>Ammonium mmr</td>
<td>Sulfate mmr</td>
<td>Ammonium mmr</td>
<td>Sulfate mmr</td>
<td>Ammonium mmr</td>
</tr>
<tr>
<td>Sulfate mmr</td>
<td>Ammonium mmr</td>
<td>Sea salt mmr</td>
<td>Ammonium mmr</td>
<td>Ammonium mmr</td>
</tr>
<tr>
<td>Secondary organic mmr</td>
<td>Aged primary organic mmr</td>
<td>Secondary organic mmr</td>
<td>Aged BC mmr</td>
<td>Aged BC mmr</td>
</tr>
<tr>
<td>Organic aerosol</td>
<td>Primary organic mmr</td>
<td>BC mmr</td>
<td>Primary organic mmr</td>
<td>BC mmr</td>
</tr>
<tr>
<td>Sea salt mmr</td>
<td>Coarse soil dust</td>
<td>Coarse sea salt</td>
<td>Coarse soil dust</td>
<td>Coarse sea salt</td>
</tr>
</tbody>
</table>

### Table 2. Organics hygroscopicity from previous measurements.

<table>
<thead>
<tr>
<th>Organics class</th>
<th>Range of (\kappa)</th>
<th>References</th>
</tr>
</thead>
<tbody>
<tr>
<td>Anthropogenic SOA</td>
<td>0.06–0.14</td>
<td>Prenni et al (2007)</td>
</tr>
</tbody>
</table>

### Table 3. Simulations performed in this study.

<table>
<thead>
<tr>
<th>Case name</th>
<th>Description</th>
</tr>
</thead>
<tbody>
<tr>
<td>CTL</td>
<td>Standard CAM, (\kappa = 0.0) (POA), (\kappa = 0.14) (SOA)</td>
</tr>
<tr>
<td>POA-hiK</td>
<td>(\kappa = 0.1) (POA), (\kappa = 0.14) (SOA)</td>
</tr>
<tr>
<td>SOA-loK</td>
<td>(\kappa = 0.0) (POA), (\kappa = 0.07) (SOA)</td>
</tr>
<tr>
<td>SOA-hiK</td>
<td>(\kappa = 0.0) (POA), (\kappa = 0.21) (SOA)</td>
</tr>
</tbody>
</table>

The total yielded mass is emitted as the SOA (gas) species. MAM then calculates condensation/evaporation of the SOA (gas) to/from aerosol modes (Aitken and accumulation mode).

3. Results

The CAM model is run for 5 years with 3-month spin-up with fixed sea surface temperatures. Anthropogenic (defined here as originating from industrial, domestic and agriculture activity sectors) emissions are from the Lamarque et al (2010) IPCC AR5 emission data set for the year 2000 (PD simulation) and the year 1850 (PI simulation). The IPCC AR5 emission data set does not provide injection heights. We used the injection heights from the AEROCOM emission protocol (Dentener et al 2006) to distribute the IPCC AR5 emissions from energy and industrial sectors at 100–300 m above the surface, and emissions from forest fire and grass fire at 0–6 km. We used the size distributions of primary emitted particles (sulfate, POA and BC) from the AEROCOM (Dentener et al 2006) to convert IPCC AR5 mass emission fluxes to number emission fluxes.
Figure 1. Annual means of column burden concentrations (mg m\(^{-2}\)) of POA in primary carbon mode (upper panel) and aged POA in the accumulation mode (middle panel), and SOA (lower panel) in the CTL run at PD (left) and at PI (right).

For POA/BC, the emitted number mean radius is 0.04 μm with a standard deviation of 1.8. The IPCC AR5 emission data set does not provide emissions for natural aerosols and precursor gases (e.g., volcanic sulfur, dimethyl sulfide (DMS)), for which we have followed the AEROCOM protocols (Dentener et al. 2006).

Figure 1 shows annual means of column burden concentrations of POA in primary carbon mode and POA aged to the accumulation mode, and SOA from the PD and PI control run. Clearly POA from primary emissions (e.g., fossil fuel, bio-fuel, and biomass burning) is aged to the accumulation mode quickly, e.g., at PD in the industrial regions where sulfate concentrations are high. This result is in general agreement with observations that carbonaceous aerosol particles are internally mixed with sulfate and other components except near the source regions (e.g., Posfai et al. 2003, Clarke et al. 2004, Moffet and Prather 2009, Wang et al. 2010). SOA column concentrations have spatial distributions similar to aged POA, and both have high concentrations in the industrial regions and in the biomass burning regions (central Africa and South America), although their seasonal variations are different (results not shown). The global spatial distributions of POA (in primary and accumulation modes) and SOA column concentrations are similar between PD and PI. However, there are several noticeable differences: aged POA concentration in the accumulation mode at PD in South and East Asia, Indonesia and in central Africa and South America is much higher than that at PI due to emission differences, while POA concentration in Eastern US is higher at PI due to more emissions from forest and grass fire and domestic heating there at PI (Lamarque et al. 2010).

The global budgets for organic aerosol (POA in primary and accumulation mode, and SOA) are given in table 4. POA in the primary mode ages to the accumulation mode with a lifetime of 1.2–1.5 days. POA in the primary mode at PI than that at PD due to significantly less sulfate available for ageing at PI (see figure 2 below). The global burden of aged POA in the accumulation mode is 4–6 times higher than that in the primary mode. Fossil and bio-fuel POA emission at PI is 36% of that at PD, however, that from biomass burning at PI is 78% of that at PD. Total POA at PI is 63% of that at PD mainly due to the lower fossil fuel and bio-fuel POA emission sources at PI. SOA (g) formation is dominated from the oxidation of biogenic VOCs (isoprene and
Figure 2 shows annual means of column burden concentration \((\text{mg m}^{-2})\) of sulfate in the control (CTL) run at PD (left) and at PI (right).

| Table 4. Global budgets for organic aerosol from the control (CTL) run at PD and PI. (Note: units are sources and sinks, Tg yr\(^{-1}\); burden, Tg; lifetime, days.) |
|-----------------|-----|-----|
|                  | PD  | PI  |
| **POA (primary mode)** |     |     |
| Sources          |     |     |
| Fossil and bio-fuel emission | 16.8 | 6.0 |
| Biomass burning emission | 32.5 | 25.2 |
| Sinks            |     |     |
| Dry deposition   | 2.9 | 2.2 |
| Wet deposition   | 0.005 | 0.003 |
| Aged to accumulation mode | 46.4 | 29.0 |
| Burden           | 0.16 | 0.13 |
| Lifetime         | 1.2 | 1.5 |
| **POA (accumulation mode)** |     |     |
| Aged from primary mode | 46.4 | 29.0 |
| Dry deposition   | 7.2 | 4.8 |
| Wet deposition   | 39.3 | 24.3 |
| Burden           | 0.95 | 0.56 |
| Lifetime         | 7.5 | 7.0 |
| **SOA**          |     |     |
| Sources          |     |     |
| Condensation of SOA (g) | 68.6 | 61.5 |
| Sinks            |     |     |
| Dry deposition   | 9.8 | 8.1 |
| Wet deposition   | 58.7 | 53.3 |
| Burden           | 1.26 | 1.13 |
| Lifetime         | 6.7 | 6.8 |

monoterpenes) in this study. Therefore we see only a slight reduction of SOA burden (by 10%) at PI compared to PD. The global SOA production \((68.6 \text{ Tg yr}^{-1}\) at PD) in this study is higher than values from the earlier studies of 11.2 Tg yr\(^{-1}\) by Chung and Seinfeld (2002), 19.1 Tg yr\(^{-1}\) by Dentener et al (2006), and 10–60 Tg yr\(^{-1}\) by Kanakidou et al (2005), and within the range of 28.9 to 75.1 Tg yr\(^{-1}\) from a more recent study by Farina et al (2010). The SOA burden of 1.26 Tg at PD in this study is higher than 0.19 Tg from Chung and Seinfeld (2002), but within the range of 0.54–2.4 Tg from Farina et al (2010).

Figure 2 shows annual means of column burden concentrations of sulfate from the PD and PI control run. Sulfate column concentrations have maximum values in the industrial regions (Asia, North America and Europe) at PD, and are significantly lower at PI. At PD, sulfate column concentrations are slightly higher than those of aged POA and SOA in the industrial regions, whereas at PI, sulfate column concentrations are much lower (e.g., in Asia and North America). The global annual burden of sulfate is 0.61 Tg S at PD (within the range of 0.53–1.07 Tg S from literature collected in Liu et al (2005)) and 0.21 Tg S at PI.

Figure 3 shows percentage changes of cloud condensation nuclei (CCN) concentration at supersaturation of 0.1% at 859 hPa between the sensitivity simulations listed in table 3 and the control (CTL) simulation at PD and at PI. Here CCN concentration diagnosed at a given supersaturation (0.1%) is used to show the impact of aerosols on cloud droplet number concentration, as the spatial distribution of changes in model predicted cloud droplet number concentration between different simulations is quite noisy due to cloud dynamical and thermodynamical feedbacks. CCN concentration at a given supersaturation is determined by a parameterization based on Köhler theory (Ghan and Zaveri 2007). The parameterization takes into consideration both the aerosol size distribution (represented by each mode) and volume weighted \(\kappa\) value from aerosol components (including organics) in each mode. For a given supersaturation and a fixed size distribution, an increase in \(\kappa\) value leads to lower critical dry size for activation, and higher CCN concentration as predicted from Köhler theory. From figure 3 at PD, an increase in \(\kappa\) value for POA from 0 to 0.1 increases CCN concentrations at 859 hPa by 20–80% over the POA source regions (see figure 1). There are also some reductions by 10–20% in the Arctic in the downwind of Siberia and Northern Canada due to more scavenging of POA in primary carbon mode when activated into cloud droplets at a higher \(\kappa\) value. A reduction of \(\kappa\) value for SOA from 0.14 to 0.07 at PD leads to reductions of CCN concentrations almost everywhere over the continents. The rainforest source regions (e.g., central Africa and South America) have reductions of 20–40%, as well as in the boreal and Arctic regions. CCN number reductions in the industrial regions are small (within 10%) because there is sufficient sulfate there to compensate aerosol hygroscopicity reduction due to SOA. An increase of \(\kappa\) value for SOA from 0.14 to 0.21 at PD has a similar impact on CCN concentrations to increasing the \(\kappa\) value for POA, although the magnitude is smaller (within 40% increase).
Figure 3 also shows the changes of CCN concentration at PI. In general the same changes in $\kappa$ value have much larger effects on CCN concentration at PI compared to those at PD. An increase in POA $\kappa$ value increases CCN concentration significantly in the industrial source regions (North America, Asia, and Europe) at PI, which is, however, much less pronounced at PD. As indicated in figure 2, POA is aged to the accumulation mode and internally mixed with sulfate and other components quickly, especially in the industrial regions (e.g., Asia) and at PD when sulfate concentrations are higher than those at PI. As a result, the variation in CCN concentration resulting from varying POA $\kappa$ value is generally small in the industrial regions at PD. This is in agreement with earlier studies that show for internally mixed particles, their hygroscopicity is dominated by inorganics such as sulfate and ammonium, and the influence of organic hygroscopicity on calculated CCN concentration is often minor there (Chang et al 2007, Prenni et al 2007, Wang et al 2008). In contrast, at PI when sulfate loading is substantially less, organics represent a larger fraction of particle volume. As a result, an increase in $\kappa$ value of POA leads to relatively larger increase in calculated CCN concentration. This disproportional change in CCN concentration can also be found with a reduction or increase in SOA $\kappa$ value.

Table 5 gives the global annual mean vertically integrated (column) CCN concentration at supersaturation of 0.1% and column cloud droplet number concentration (CDNUMC) for all four runs listed in table 3 at PD and PI. Clearly there are larger changes in both CCN and CDNUMC concentrations at PI compared to PD as $\kappa$ values vary. We note that global mean CCN column concentration at supersaturation of 0.1% is substantially higher than CDNUMC. This is because CCN concentration at supersaturation of 0.1% is calculated for the three-dimensional aerosol fields predicted in the model, whereas model predicted cloud droplet number concentration is limited to the cloudy regions where cloud droplets exist.

Figure 4 shows annual zonal mean changes of CDNUMC, shortwave cloud forcing (SWCF), liquid water path (LWP) and low-level cloud cover (CLDLOW) between PD and PI for the four simulations listed in table 3. We note that cloud response
Figure 4. Annual zonal mean changes of column cloud droplet number concentration (CDNUMC), shortwave cloud forcing (SWCF), liquid water path (LWP) and low-level cloud cover (CLDLOW) between PD and PI for the four simulations.

Table 5. Global annual mean vertically integrated (column) cloud condensational nuclei (CCN) concentration (×10^{10} m^{-2}) at supersaturation of 0.1% and cloud droplet number concentration (CDNUMC, ×10^9 m^{-2}) for all four runs at PD and PI.

<table>
<thead>
<tr>
<th>Case name</th>
<th>CTL</th>
<th>POA-hiK</th>
<th>SOA-loK</th>
<th>SOA-hiK</th>
</tr>
</thead>
<tbody>
<tr>
<td>Column CCN*</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>PD</td>
<td>25.6</td>
<td>27.6(+7.8%)</td>
<td>23.7(−7.4%)</td>
<td>27.3(+6.6%)</td>
</tr>
<tr>
<td>PI</td>
<td>14.9</td>
<td>16.7(+12.1%)</td>
<td>12.9(−13.4%)</td>
<td>16.6(+11.4%)</td>
</tr>
<tr>
<td>CDNUMC*</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>PD</td>
<td>14.8</td>
<td>15.0(+1.3%)</td>
<td>14.3(−3.4%)</td>
<td>15.1(+2.1%)</td>
</tr>
<tr>
<td>PI</td>
<td>9.9</td>
<td>10.4(+5.1%)</td>
<td>9.1(−8.1%)</td>
<td>10.3(+4.0%)</td>
</tr>
</tbody>
</table>

* The values in parentheses are for the percentage changes of column CCN concentration and CDNUMC between the sensitivity runs (POA-hiK, SOA-loK, and SOA-hiK) and the control run (CTL) for both PD and PI.

The global, NH and SH annual mean changes between PD and PI of SWCF, LWP, CLDLOW and CDNUMC for different runs are listed in Table 6. AIF is defined here as SWCF change between PD and PI as LWCF change is much smaller (0.1 W m^{-2}). We can see that AIF has the largest difference between POA-hiK and SOA-loK runs (by 0.4 W m^{-2} on the global mean) along with largest change in ΔLWP between these two runs (by 1 g m^{-2} on the global mean), in consistence with the smallest increase in POA-hiK (4.6 × 10^9 m^{-2} on the global mean) and the largest increase of CDNUMC between PD and PI in SOA-loK (5.2 × 10^9 m^{-2} on the global mean). Changes between PD and PI (△SWCF, △LWP, △CLDLOW and △CDNUMC) for all four runs occur mainly in the NH due to more anthropogenic emissions. Differences of △SWCF,
Table 6. Global, northern hemisphere (NH) and southern hemisphere (SH) annual mean changes (Δ) of shortwave cloud forcing (SWCF, W m\(^{-2}\)), liquid water path (LWP, g m\(^{-2}\)), low-level cloud cover (CLDLOW, %), and column droplet number concentration (CDNUMC, ×10\(^3\) m\(^{-2}\)) between PD and PI (PD–PI) for all runs.

<table>
<thead>
<tr>
<th>Case name</th>
<th>CTL(\Delta)</th>
<th>POA-hiK</th>
<th>SOA-loK</th>
<th>SOA-hiK</th>
</tr>
</thead>
<tbody>
<tr>
<td>ΔSWCF Global</td>
<td>−1.3</td>
<td>−1.1</td>
<td>−1.5</td>
<td>−1.2</td>
</tr>
<tr>
<td>NH</td>
<td>−2.2</td>
<td>−1.9</td>
<td>−2.5</td>
<td>−2.0</td>
</tr>
<tr>
<td>SH</td>
<td>−0.3</td>
<td>−0.3</td>
<td>−0.4</td>
<td>−0.4</td>
</tr>
<tr>
<td>ΔLWP Global</td>
<td>3.9</td>
<td>3.6</td>
<td>4.6</td>
<td>3.8</td>
</tr>
<tr>
<td>NH</td>
<td>6.3</td>
<td>5.7</td>
<td>7.2</td>
<td>6.0</td>
</tr>
<tr>
<td>SH</td>
<td>1.5</td>
<td>1.5</td>
<td>2.0</td>
<td>1.6</td>
</tr>
<tr>
<td>ΔCLDLOW Global</td>
<td>0.3</td>
<td>0.2</td>
<td>0.3</td>
<td>0.3</td>
</tr>
<tr>
<td>NH</td>
<td>0.6</td>
<td>0.5</td>
<td>0.7</td>
<td>0.6</td>
</tr>
<tr>
<td>SH</td>
<td>0.0</td>
<td>−0.1</td>
<td>−0.1</td>
<td>0.0</td>
</tr>
<tr>
<td>ΔCDNUMC Global</td>
<td>4.9</td>
<td>4.6</td>
<td>5.2</td>
<td>4.8</td>
</tr>
<tr>
<td>NH</td>
<td>8.0</td>
<td>7.4</td>
<td>8.3</td>
<td>7.7</td>
</tr>
<tr>
<td>SH</td>
<td>1.8</td>
<td>1.8</td>
<td>2.1</td>
<td>1.9</td>
</tr>
</tbody>
</table>

\(\Delta\) Global annual mean SWCF, LWP, CLDLOW, and CDNUMC of the control run at PD are −47.1 W m\(^{-2}\), 45.6 g m\(^{-2}\), 37.1%, and 14.8 × 10\(^3\) m\(^{-2}\), respectively.

The impact of POA to the budget of organic aerosol, further constraint of their hygroscopicity in global models will help to reduce the uncertainty in simulated AIF.

An implicit assumption in the simulations is that for both organics classes, their hygroscopicities do not change from PI to PD. Anthropogenic emissions can strongly influence ozone and nitrogen oxide (NO\(_x\)) concentrations, which have been shown to impact the formation of secondary organic aerosols (Kanakidou et al 2005, Presto et al 2005, Dommen et al 2006, Kroll et al 2006, Ng et al 2007, Carlton et al 2009). Studies also suggest the VOC to NO\(_x\) ratio may influence the composition of SOA formed from biogenic VOC and therefore potentially its \(\kappa\) value. A recent study finds little variation in the \(\kappa\) value of SOA formed from isoprene photo-oxidation at different VOC to NO\(_x\) ratios (King et al 2010). However, there have been essentially no studies on the impact of NO\(_x\) on the \(\kappa\) value of SOA formed from other biogenic precursors. While the SOA \(\kappa\) value may have relatively small impact on calculated CCN concentration for PD due to the internal mixing with high-hygroscopic inorganic species, this change of SOA \(\kappa\) value from PI to PD as a result of anthropogenic influence may lead to additional uncertainty in simulated AIF.

The biomass burning organics and biogenic secondary organics dominate the budgets of POA and SOA, especially for PI. Currently, the experimentally determined \(\kappa\) values of SOA and biomass burning POA show a range of 0.06 to −0.3. Given the importance of understanding the CCN concentration at PI, the \(\kappa\) values of SOA and biomass burning POA need to be better understood and represented in models to reduce the uncertainties in AIF. As fossil fuel POA and biomass burning POA have quite different hygroscopicities, separating the two types of POA in future model development would improve the simulation of AIF. Furthermore, ageing processes in the atmosphere can also lead to the increase of hygroscopicity of POA and SOA. Field studies suggest that the \(\kappa\) value of SOA increases from 0 to −0.2 during ageing in the atmosphere (Jimenez et al 2009), a range that is even greater than that simulated in the current study. The representation of this change of \(\kappa\) due to atmospheric ageing is also an important step for improving the simulation of AIF. Besides organics hygroscopicity, large uncertainties also exist on the SOA formation simulated in global models. The impact of organics formation on AIF will be examined in future studies.

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